

Title	Extraordinary wavelength dispersion of birefringence in cellulose triacetate film with anisotropic nanopores
Author(s)	Nobukawa, Shogo; Shimada, Hikaru; Aoki, Yoshihiko; Miyagawa, Azusa; Doan, Vu Ahn; Yoshimura, Hiroshi; Tachikawa, Yutaka; Yamaguchi, Masayuki
Citation	Polymer, 55(15): 3247-3253
Issue Date	2014-05-22
Type	Journal Article
Text version	author
URL	http://hdl.handle.net/10119/13504
Rights	Copyright (C)2014, Elsevier. Licensed under the Creative Commons Attribution-NonCommercial-NoDerivatives 4.0 International license (CC BY-NC-ND 4.0). [http://creativecommons.org/licenses/by-nc-nd/4.0/] NOTICE: This is the author's version of a work accepted for publication by Elsevier. Shogo Nobukawa, Hikaru Shimada, Yoshihiko Aoki, Azusa Miyagawa, Vu Ahn Doan, Hiroshi Yoshimura, Yutaka Tachikawa, Masayuki Yamaguchi, Polymer, 55(15), 2014, 3247-3253, http://dx.doi.org/10.1016/j.polymer.2014.05.037
Description	

1 Extraordinary wavelength dispersion of birefringence in cellulose triacetate film with
2 anisotropic nanopores

3

4 Shogo NOBUKAWA^{1*}, Hikaru SHIMADA¹, Yoshihiko AOKI¹, Azusa MIYAGAWA¹, Vu Ahn
5 DOAN^{1†}, Hiroshi YOSHIMURA², Yutaka TACHIKAWA³, and Masayuki YAMAGUCHI^{1**}

6

7 ¹*School of Materials Science, Japan Advanced Institute of Science and Technology, 1-1 Asahidai,*
8 *Nomi, Ishikawa 923-1292, Japan*

9 ²*DIC Corporation, Chiba Plant, 12 Yawata-kaigandori, Ichihara, Chiba 290-8585, Japan*

10 ³*DIC Corporation, Central Research Laboratories, 631 Sakado, Sakura, Chiba 285-8668, Japan*

11

12 *Corresponding author:

13 Tel: +81-761-51-1626

14 School of Materials Science, Japan Advanced Institute of Science and Technology, 1-1 Asahidai,

15 Nomi, Ishikawa 923-1292, Japan

16 E-mail: nobukawa@jaist.ac.jp (S. N.)

17

18 **Corresponding author:

19 Tel: +81-761-51-1621

20 School of Materials Science, Japan Advanced Institute of Science and Technology, 1-1 Asahidai,

21 Nomi, Ishikawa 923-1292, Japan

22 E-mail: m_yama@jaist.ac.jp (M. Y.)

23

24 †Present address: Hanoi University of Science and Technology, 1 Dai Co Viet Road, Hanoi,

25 Vietnam

26

1 **ABSTRACT**

2 We examined birefringence in a stretched film of cellulose triacetate (CTA) after extraction
3 of an immiscible component. The CTA film plasticized by di(2-ethylhexyl) adipate (DOA), which
4 was added as the immiscible additive, exhibited negative birefringence to the same degree as the
5 pure CTA film. Following removal of DOA from the film by immersion into methanol, the
6 birefringence of the blend film changed dramatically from negative to positive. Moreover, the
7 wavelength dependence also changed from ordinary to extraordinary, in which the absolute value of
8 birefringence increases with wavelength. Scanning electron microscope (SEM) images revealed
9 nanoscale ellipsoidal pores in the film after the extraction, suggesting that DOA was segregated and
10 formed ellipsoidal domains in the CTA matrix during annealing and stretching. According to an
11 optical theory for the nanoporous structure, we found that the form birefringence contributes to
12 control of the optical properties of the CTA film. This phenomenon could be utilized in the design
13 of high-performance optical films, such as quarter waveplate, because sign and wavelength
14 dispersion of birefringence can be controlled even for a single component film.

15

16 **Keywords:** Nano-porous structure, Cellulose triacetate, Birefringence

17

18

1 INTRODUCTION

2 Cellulose esters for use in optical films have been studied because they have favorable
3 optical properties including transparency, thermostability, chemical resistance, and dimensional
4 stability [1-6]. In particular, birefringence (Δn), which is defined as the difference between two
5 refractive indices, is one of the most important properties of optical films such as retardation and
6 protection films [6-8]. Modern high-performance display devices, e.g., three-dimensional (3D) and
7 organic electro luminescence (EL) displays, require a quarter waveplate with well-controlled Δn and
8 wavelength dispersion [6, 7].

9 For a quarter waveplate, the proportionality relationship between Δn and wavelength is
10 important. To improve the optical properties of films, copolymerization [7, 9], blending with other
11 polymers or with additives [10-16], and lamination [17] have to be used because the wavelength
12 dispersion is determined only by the chemical structure of the polymers. However, these methods
13 exhibit some problems; copolymerization reduces thermo-resistance, and polymer blending with low
14 compatibility decreases film transparency. With lamination, the thermal expansion mismatch
15 between polymer sheets restricts the useful temperature range.

16 The birefringence of polymeric materials has three components given by:

17

18

$$\Delta n = \Delta n_O + \Delta n_G + \Delta n_F \quad (1)$$

19

1 Here, Δn_O , Δn_G , and Δn_F are orientation, glassy and form birefringence, respectively. The first
2 term pertains to the chain orientation and the second is related to the distortion of chemical structure
3 due to an applied force. The third term is associated with anisotropic nanostructures such as
4 ellipsoidal structures. Based on an anisotropic dielectric theory [18] for the form birefringence,
5 Ibn-Elhaj and Schadt [19] improved the birefringence of polymer thin films by the anisotropic
6 nano-corrugated surface pattern. Therefore, the nano-porous structure possibly modifies
7 birefringence of the polymeric films.

8 Moreover, many studies investigated porous structures in polymeric films for the application
9 for separation membranes and separators in lithium ion second cells [20-22]. In general, porous
10 films are prepared by using a thermo-induced phase separation (TIPS) of the matrix polymers and
11 small amount of immiscible components, which are removed by immersing the films into selective
12 solvents. We think that the porous structure by the TIPS technique contributes to the optical
13 anisotropy, i.e., the form birefringence in cellulose ester films.

14 Because Δn_G in equation (1) is negligibly small compared to other terms for the
15 hot-stretched polymer films, the birefringence property of stretched cellulose ester films can be
16 modified by incorporation of an anisotropic nano-structure. In this study, the optical birefringence
17 and its wavelength dispersion of cellulose triacetate (CTA, Figure 1), used as a model polymer, was
18 improved with anisotropic nano-pores.

19

1 [Figure 1]

2
3 In order to design nano-porous structure such as sea-island structures in CTA films, the
4 phase separation technique can be utilized. Since CTA is a crystalline polymer that is insoluble in
5 most solvents, such as methanol, removal of low-mass molecules (LMs) from the sea-island structure
6 can be accomplished using the immersion method. Based on this idea, di(2-ethylhexyl) adipate
7 (DOA, Figure 1) was used as the immiscible LM and the pores were formed from sea-island
8 structures resulting from phase separation of the CTA/DOA blend. In order to investigate the form
9 birefringence, effects of chain orientation, crystallinity, and moisture content on birefringence of
10 CTA film were also discussed.

11 12 **EXPERIMENTAL**

13 **Samples**

14 The CTA used in this study was produced by the Daicel Corporation (Japan). The degree
15 of acetylation per pyranose unit of CTA was 2.96, as determined by $^1\text{H-NMR}$ spectroscopy. The
16 weight-average and number-average molecular weights (M_w and M_n) of CTA were 3.5×10^5 and
17 1.3×10^5 , respectively, as determined by gel permeation chromatography (GPC, HLC-8020 Tosoh,
18 Japan) with TSK-GEL[®] GMHXL compared with polystyrene standards. DOA was used as a
19 plasticizer to prepare the domains. DOA is commercially available and was supplied by Daihachi

1 Chemical Industry (Japan).

2 CTA and CTA/DOA films were prepared using a solution-casting method; melt processing
3 is not applicable because thermal degradation takes place near the melting point of CTA (around
4 300 °C). CTA and DOA in a ratio of 10/1 wt/wt were dissolved in a binary solvent of
5 dichloromethane/methanol (9/1 vol/vol) and a 3 wt% solution was prepared. The solvent was
6 evaporated at room temperature (RT) for 1 day at ordinary pressure and dried *in vacuo* for 1 day
7 without controlling humidity. Since DOA is not volatile, the composition of the blend film was
8 controlled by the ratio of components. The film thickness was 100–150 μm.

9

10

[Table 1]

11

12 **Measurements**

13 Dynamic mechanical analysis (DMA) of the sample films was performed to measure storage
14 and loss moduli (E' and E'' , respectively) at 10 Hz as a function of temperature using a tensile
15 oscillatory rheometer (DVE-E4000, UBM, Japan) from 0 to 250 °C at a heating rate of 2 °C min⁻¹.
16 The glass transition temperature (T_g) was estimated from a peak of E'' in the DMA data and is shown
17 in Table 1.

18 A hot-stretching test for the films was carried out with a strain rate of 0.05 s⁻¹ using a tensile
19 drawing machine (DVE-3, UBM, Japan). To avoid the thermal history effect, the films were annealed

1 for 10 min before stretching. The drawing temperatures (T_{draw}) were adjusted to obtain the same
2 stress levels for CTA and CTA/DOA films, as shown in Table 1. The films were immediately
3 quenched by cold air blowing after stretching to avoid relaxation of molecular orientation.

4 The orientation birefringence (Δn) of the films was measured as a function of wavelength
5 using an optical birefringence analyzer (KOBRA-WPR, Oji Scientific Instruments, Japan). The
6 details of the optical system have been described previously [23]. Refractive indices at various
7 wavelengths of CTA and DOA were measured by using an Abbe refractometer (DR-M2, Atago,
8 Japan).

9 To remove DOA, the stretched and unstretched films were immersed in methanol for at least
10 8 hours and dried *in vacuo* for 1 day at RT. The complete removal of DOA from the film was
11 confirmed by the $^1\text{H-NMR}$ spectra before/after immersion using an AVANCE III NMR 400 (Bruker
12 Corporation, Germany). The obtained films were kept in a humidity chamber (IG420, Yamato,
13 Japan) at 25 °C and 50% relative humidity (RH) for 1 day to avoid the effects of uncontrolled
14 moisture on the optical properties, as previously reported [24]. The moisture content of the films
15 was determined using a Karl Fisher moisture meter (Mitsubishi Chemical Analytec Co. Ltd, Japan).

16 The morphology of the films was examined using a scanning electron microscope (SEM)
17 (S4100, Hitachi Ltd, Japan) with an acceleration voltage of 20 kV. Prior to the observation, the
18 surfaces of the films were coated with Pt-Pd in an ion-sputter machine (E1010, Hitachi Ltd, Japan).
19 In order to examine the sea-island structure in CTA/DOA films, a light scattering (LS) measurement

1 was carried out at room temperature by using a LS analyzer (DYNA-3000, Otsuka Electronics Co.,
2 Ltd., Japan). The correlation length, which reflects the morphology of CTA/DOA films, was
3 estimated from the LS data.

4 The wide-angle X-ray diffraction (WAXD) pattern was measured at RT using a powder
5 X-ray diffractometer (RINT-2500, Rigaku, Japan) by refractive mode to evaluate the crystallinity of
6 CTA in the films. The experiments were carried out using CuK α radiation at 40 kV and 30 mA at a
7 scanning rate of 1° min⁻¹ over 2 hours (Bragg angle range from 5 to 60°).

8

9 **RESULTS and DISCUSSION**

10 **Plasticization effect of DOA on dynamic mechanical property of CTA**

11 Dynamic mechanical properties of polymeric materials represent miscibility or
12 immiscibility of the blends. Furthermore, the properties are also an important consideration when
13 preparing a stretched film by melt processing because the applied stress affects the chain orientation
14 of the polymer. Figure 2 shows the temperature dependence of E' and E'' for CTA and CTA/DOA
15 films. The glass-to-rubber transition of the blend shifted to a lower temperature and broadened
16 compared with that of the bulk, implying that DOA acts as a plasticizer for CTA. Yamaguchi et al.
17 [25] reported that DOA plasticizes cellulose acetate propionate (CAP), an analog of CTA. The
18 addition of DOA was reported to decrease the T_g of CAP by 35 °C compared with only 5 °C for CTA,
19 suggesting that the solubility of DOA and CTA is lower than that of DOA and CAP. Therefore, DOA

1 is possibly immiscible with CTA on the molecular level, although the blend film was transparent
2 after solution casting.

3 The rubbery plateau is also observed for general uncross-linked polymers owing to chain
4 entanglement, and the value is at least 0.1 to 1 MPa. In Figure 2, E' curves for CTA and CTA/DOA
5 also represent the rubbery plateau beyond 200 °C. However, the plateau modulus is over 10 MPa,
6 which is much larger than the general value. This suggests that microcrystallites of CTA act as
7 cross-link points. Furthermore, the plateaus in both curves represent the same level, meaning that the
8 cross-link densities of CTA are comparable in bulk and blend. If the crystallite size is not affected by
9 the presence of DOA, the degrees of crystallinity for CTA are similar in both systems.

10

11

[Figure 2]

12

13 **Orientation birefringence of CTA and CTA/DOA**

14 Solution-cast films of CTA and CTA/DOA were uniaxially stretched with a draw ratio of 1.5
15 at T_{draw} , as shown in Table 1. Stress–strain curves were obtained during stretching, as shown in
16 Figure 3. The figure demonstrates the same stress level for the two stretched films, indicating that the
17 chain orientations of CTA in bulk and blend are similar, as discussed later.

18

19

[Figure 3]

1

2 When polymeric materials are uniaxially stretched beyond T_g , birefringence (Δn) is
 3 generated owing to the chain orientation, as represented by:

4

$$5 \quad \Delta n = \Delta n^0 F \quad (2)$$

6

7 Here, Δn^0 is the intrinsic birefringence reflecting the polarizability anisotropy of the chain segment
 8 for the polymers, and F is the orientation function representing the chain alignment to the stretching
 9 direction. Since stress (σ), as well as birefringence, is generated, the following relationship is
 10 obtained:

11

$$12 \quad \Delta n = C \sigma \quad (3)$$

13

14 Here, C is a stress-optical coefficient, which is related to the polarizability anisotropy of the polymer
 15 chain. This equation is known as the stress-optical rule (SOR). From equations (2) and (3), the result
 16 in Figure 2 suggests that the chain orientation of CTA is the same in bulk and blend.

17

For polymer/small molecule blends, Δn is the sum of two components, as given by:

18

$$19 \quad \Delta n = \phi_{poly} \Delta n_{poly} + (1 - \phi_{poly}) \Delta n_{LM} \quad (4)$$

20

1 where ϕ_{poly} is the volume fraction of the polymer, and Δn_{poly} and Δn_{LM} are the birefringence values of
 2 the two components in the pure state. However, the dynamic mechanical data in Figure 2 indicate
 3 that CTA is not completely miscible with DOA; that is, phase separation occurs. Since the CTA/DOA
 4 film seems to be transparent, the separated domain of DOA might be smaller than wavelength of
 5 visible lights (at least 400 nm), which is related to the Mie scattering. Therefore, the form
 6 birefringence (Δn_F) caused by the separated domain should be considered, as represented by:

$$\Delta n = \phi'_{CTA} \Delta n_{CTA} + (1 - \phi'_{CTA}) \Delta n_{DOA} + \Delta n_F \quad (5)$$

7
 8
 9
 10 The prime mark in ϕ'_{CTA} represents the separated CTA-rich phase in the blend system. Figure 4
 11 shows the wavelength dependence of orientation birefringence (Δn) of CTA and CTA/DOA films
 12 after stretching. As seen in the figure, Δn data for the two films are comparably negative, and both
 13 curves exhibit ordinary wavelength dependence, that is, the absolute value decreases with
 14 wavelength increasing. The result means that the second and third terms on the right-hand side of
 15 equation (5) are negligibly small. The form birefringence Δn_F is observed in phase-separated
 16 systems composed of two or three phases having different refractive indices. Actually, since the
 17 refractive indices (n) of CTA and DOA are almost similar (1.48 and 1.44 at 589 nm, respectively),
 18 Δn_F can be negligible as explained later.

19

[Figure 4]

Abd Manaf et al. [14] found that the addition of tricresyl phosphate (TCP) changed the Δn of CTA from negative to positive because the orientation of TCP was affected by intermolecular nematic interaction (NI) with the matrix polymer. Using their birefringence data from polymer/aromatic molecule blends, Nobukawa et al. [15, 16] reported that the rigid and anisotropic structure of the LM enhances the NI strength. As shown in Figure 4, the second term, $(1-\phi'_{CTA})\Delta n_{DOA}$, which is originated from DOA, is small. There are two possible explanations for this. The first possibility, from the above discussion, is the absence of NI between CTA and DOA, that is, the orientation of DOA is almost zero. The second possible influence is that the CTA matrix phase does not contain DOA because of phase separation as discuss below.

Phase separation in stretched CTA/DOA film

In order to investigate the phase separation in stretched CTA/DOA film, a static light scattering (LS) experiment was performed at room temperature. According to the statistical approaches by Debye and Bueche [26], the scattered intensity of depolarized light, I , is given as a function of scattered vector q and the correlation length ξ ,

$$I = \frac{A}{(1 + \xi^2 q^2)^2} \quad (6)$$

1

2 Here, A is a constant depending on the incident light. The Debye-Bueche (DB) type plots ($I^{-1/2}$ vs.

3 q^2) represents a linear relation [27] as,

4

$$5 \quad I^{-1/2} = \frac{1}{A^{1/2}} + \frac{\xi^2}{A^{1/2}} q^2 \quad (7)$$

6

7 Figure 5(A) shows the DB plot for CTA/DOA film after stretching, which indicates the sea-island

8 type phase separation. By applying equation (7), the correlation length ξ was estimated to be 0.61

9 $\pm 0.08 \mu\text{m}$ representing the distance between the separated islands of DOA.

10

11

[Figure 5]

12

13 **Birefringence and porous structure in CTA/DOA film after extraction of DOA**

14 The stretched CTA/DOA film was immersed into methanol for a half day and dried in *vacuo*,

15 to remove DOA from the separated islands. As a control, CTA films were also immersed.

16 The $^1\text{H-NMR}$ spectra, which are shown in supporting information, indicated the complete extraction

17 of DOA after the immersion. Figure 4 also shows the wavelength dispersion of Δn for CTA and

18 CTA/DOA after immersion. The change in Δn for CTA was zero within experimental error, meaning

19 that the chain orientation of CTA did not relax and the degree of crystallinity was held even after

1 immersion. In contrast, the Δn of CTA/DOA dramatically changed from negative to positive after
2 removal of DOA. Since nanoscale DOA domains existed in the CTA/DOA before immersion,
3 nanoscale pores might appear. The LS data of CTA/DOA film after the extraction, which is shown
4 in Figure 5(B), exhibits the similar ξ ($=0.63 \pm 0.13 \mu\text{m}$) to the film before extraction, suggesting that
5 the nano pores are still maintained in CTA film. In order to investigate the morphology of the film,
6 images were taken using scanning electron microscopy (SEM).

7 Figure 6 compares the SEM images for unstretched, annealed and stretched CTA/DOA films
8 after immersion. The unstretched film before annealing shown in Figure 6(A) was not subject to
9 phase separation because no structures such as pores were observed in the surface or cross-section. In
10 Figure 6(B), pores are visible in the cross-section of the unstretched films following the extraction.
11 After stretching, ellipsoidal pores appeared, as shown in Figure 6(C). By assuming the affine
12 deformation, the aspect ratio of DOA domain should be 1.5/0.8 (long axis/short axis), which is
13 consistent with that of the ellipsoidal pore in Figure 6(C). Therefore, the SEM images suggest that
14 DOA domains appear because of annealing above T_g , and the spherical domains are deformed into
15 needle shapes by stretching.

16

17

[Figure 6]

18

19 **Effects of moisture and crystallinity on birefringence**

1 The effects of moisture and crystallinity on total birefringence of CTA film were confirmed
 2 as below. Abd Manaf et al. [24] investigated the effect of moisture on the Δn of CTA by controlling
 3 the humidity. According to them, 1 wt% water content increases the Δn of CTA by -2×10^{-4} . As
 4 seen in Table 2, the water contents in CTA and CTA/DOA films do not change after the immersion in
 5 methanol. Hence, the moisture effect is negligible for Δn change, as shown in Figure 4.

6 In general, crystalline and amorphous regions in crystalline and semi-crystalline polymers,
 7 including CTA, contribute to orientation birefringence, as represented by the following equation:

$$\Delta n = x_c \Delta n_c^0 F_c + (1 - x_c) \Delta n_a^0 F_a \quad (8)$$

8

9 Here, x_c is the degree of crystallinity. The subscripts c and a represent crystalline and amorphous
 10 regions, respectively. For generally crystalline polymers, such as poly(ethylene 2,6-naphthalate), the
 11 two intrinsic birefringences, Δn_c^0 and Δn_a^0 , are not the same [28]. Therefore, the degree of
 12 crystallinity changes the total birefringence.
 13
 14

15 Figure 7 shows the X-ray diffraction (XRD) patterns for CTA and CTA/DOA films scanned
 16 under various conditions. The unstretched films for both systems showed broad peaks, which
 17 agrees with data reported by Cao et al. [29]. However, stretched and immersed films showed strong
 18 peaks reflecting crystalline structure and implying that the crystallization was induced by the hot
 19 stretching. According to equation (6), the Δn of CTA should be discussed by considering the two
 20 components from crystalline and amorphous regions. However, since the XRD patterns for

1 stretched and immersed films were similar, the degree of crystallinity was not changed by the
2 stretching process. Hence, the effects of crystallinity and moisture content on Δn can be ignored.

3

4

[Figure 7]

5

6 What is the reason for the birefringence change? As represented by equation (5), the form
7 birefringence (Δn_F) can be found in separate phases with very different refractive indices (n 's).
8 Moreover, since the porous structure that appeared after the extraction of DOA is filled with air, the
9 difference in refractive index is much larger than that between CTA and DOA. Based on this idea,
10 the form birefringence observed in CTA/DOA films is discussed in the next section.

11

12 **Form birefringence in anisotropic porous CTA.**

13 As can be seen in Figure 6(C), the stretched CTA/DOA film after the extraction of DOA had
14 an anisotropic porous structure. Figure 8 is a simplified depiction of the anisotropic porous
15 structure of the stretched CTA blend film after removal of DOA by the extraction method.
16 Definitions of D and f are represented in the figure.

17

18

[Figure 8]

19

1 Ibn-Elhaj and Schadt [19] examined the contribution of anisotropic nano-corrugated pattern
 2 to the optical birefringence of thin polymeric films, based on the anisotropic dielectric theory.
 3 However, the theory was developed by utilizing a zero-order approximation where electric field and
 4 displacement vector are constant [30]. Therefore, Richter et al. [31] determined the form
 5 birefringence (Δn_F) in the grating system with anisotropic microstructures by applying more precise
 6 second-order solution given by,

$$\Delta n_F = n_{2,\parallel} - n_{2,\perp} \quad (9)$$

$$n_{0,\parallel} = [(1-f)n_{poly}^2 + fn_{dm}^2]^{1/2}$$

$$n_{0,\perp} = \left[\frac{n^2 n_0^2}{(1-f)n_{poly}^2 + fn_{dm}^2} \right]^{1/2} \quad (10)$$

$$n_{2,\parallel} = \left[n_{0,\parallel}^2 + \frac{1}{3} \left(\frac{D}{\lambda} \right)^2 \pi^2 f^2 (1-f^2) (n_{poly}^2 - n_{dm}^2)^2 \right]^{1/2}$$

$$n_{2,\perp} = \left[n_{0,\perp}^2 + \frac{1}{3} \left(\frac{D}{\lambda} \right)^2 \pi^2 f^2 (1-f^2) \left(\frac{1}{n_{poly}^2} - \frac{1}{n_{dm}^2} \right)^2 n_{0,\parallel}^6 n_{0,\perp}^2 \right]^{1/2} \quad (11)$$

11
 12 Here, the number in subscript following n represents the approximation order for the equation
 13 solution, and λ is the light wavelength. The refractive indices of polymer and pore (or domain) are
 14 denoted by n_{poly} and n_{dm} , respectively.

15 From equations (9)–(11), the third term in equation (5) can be calculated for CTA/DOA
 16 (before extraction) and porous CTA (after extraction) films. To fit the calculated result to the

1 experimental data for CTA/pore film in Figure 4, f was determined to be 0.005, and D was $0.6 \mu\text{m}$ as
2 estimated from the LS data. These values seem to be consistent with the SEM image. However, f
3 value is not consistent with the concentration of DOA ($= 0.09$), suggesting that the pore size will be
4 reduced owing to the osmotic pressure difference between the CTA and methanol phases during the
5 extraction. The calculated results are shown in Figure 9. The refractive indices at various
6 wavelengths for CTA and DOA are shown in the figure while that for pore is 1.0. As shown in the
7 figure, the contribution of Δn_F in CTA/DOA film is smaller than 10^{-6} while that in the CTA/pore film
8 is 15×10^{-4} . In addition, the weaker wavelength dependence of the calculated Δn_F agrees with the
9 change of Δn from the CTA/DOA and CTA/pore films represented in Figure 4.

10

11

[Figure 9]

12

13 We conclude that the form birefringence Δn_F , which is generated by the extraction of
14 immiscible plasticizer, may change total birefringence of CTA film from negative to positive.
15 Furthermore, since Δn_F and the orientation birefringence Δn_O exhibit weak and strong
16 wavelength dependences, respectively, the total birefringence ($= \Delta n_F + \Delta n_O$) shows extraordinary
17 dependence.

18 In order to control the total birefringence of CTA/pore film, both Δn_O and Δn_F should be
19 considered. The former can be controlled by the experimental conditions such as the stretching

1 ratio and temperature as reported in many papers. In contrast, the latter has not been investigated.
2 The theory represented by equations (9)-(11) does not predict the stretching ratio dependence of
3 Δn_F . We think that these stretching conditions will change size and aspect ratio of the pore. In
4 the future work, the relationship between the pore size and Δn_F should be investigated by varying
5 the preparation conditions of the nanoporous film.

6

7 **Conclusions**

8 We investigated orientation birefringence in a nanoporous film of cellulose triacetate (CTA),
9 which was prepared from CTA/di(2-ethylhexyl) adipate (DOA) by immersion into methanol to
10 extract DOA. Before the extraction, although the CTA/DOA film showed phase separation due to
11 low miscibility, the orientation birefringence of CTA/DOA was negative and similar to that of CTA,
12 implying that the DOA domains make no contribution to the birefringence. After the extraction, the
13 birefringence of CTA became positive. Based on the results of moisture content, XRD and SEM, we
14 think that the anisotropic porous structure of the CTA film after extraction is responsible for the
15 birefringence change.

16 The form birefringence of the anisotropic microstructure was estimated according to surface
17 gratings theory. As a result, agreement between theoretical and experimental results was achieved.
18 We suggest that, even in films comprising a single component, the optical birefringence property can
19 be modified by anisotropic nanopores created by the immersion method.

1

2 **Acknowledgement**

3 This work was partly supported by a Grant-in-Aid for Young Scientists B (25870268), and
4 by a grant from the Kyoto Technoscience Center in 2013. The authors would like to thank Otsuka
5 Electronics Co., Ltd. for the measurement of light scattering.

6

1 **REFERENCES**

- 2 1. Edgar KJ, Buchanan CM, Debenham JS, Rundquist PA, Seiler BD, Shelton MC, and Tindall
3 D. *Prog. Polym. Sci.* 2001;26:1605-1688.
- 4 2. Law PW, Longdon A, and Willins GG. *Macromol. Symp.* 2004;208:293-322.
- 5 3. Ohno T and Nishio Y. *Macromolecules* 2007;40:3468-3476.
- 6 4. Maeda A and Inoue T. *Nihon Reoroji Gakkaishi* 2011;39:159-163.
- 7 5. Songsurang K, Miyagawa A, Abd Manaf ME, Phulkerd P, Nobukawa S, and Yamaguchi M.
8 *Cellulose* 2012;20:83-96.
- 9 6. Yamaguchi M, Abd Manaf ME, Songsurang K, and Nobukawa S. *Cellulose*
10 2012;19:601-613.
- 11 7. Uchiyama A, Ono Y, Ikeda Y, Shuto H, and Yahata K. *Polym. J.* 2012;44:995-1008.
- 12 8. Tagaya A and Koike Y. *Polymer J.* 2012;44:306-314.
- 13 9. Tagaya A, Ohkita H, Harada T, Ishibashi K, and Koike Y. *Macromolecules*
14 2006;39:3019-3023.
- 15 10. Saito H and Inoue T. *J. Poly. Sci. Part-B Poly. Phys.* 1987;25:1629-1636.
- 16 11. Tagaya A, Iwata S, Kawanami E, Tsukahara H, and Koike Y. *App. Opt.* 2001;40:3677-3683.
- 17 12. Uchiyama A and Yatabe T. *Japan J. App. Phys. Part-1* 2003;42:3503-3507.
- 18 13. Rao YQ, Chen JL, and Ishikawa T. Control of Birefringence Dispersion Utilizing
19 Nanocomposites. In: Company EK, editor. US Patent, vol. US 2007/0042137 A1. United
20 State, 2005.
- 21 14. Abd Manaf ME, Tsuji M, Shiroyama Y, and Yamaguchi M. *Macromolecules*
22 2011;44:3942-3949.
- 23 15. Nobukawa S, Urakawa O, Shikata T, and Inoue T. *Macromolecules* 2010;43:6099-6105.
- 24 16. Nobukawa S, Aoki Y, Yoshimura H, Tachikawa Y, and Yamaguchi M. *J. Appl. Polym. Sci.*
25 2013;130:3465-3472.
- 26 17. Cho CK, Kim JD, Cho K, Park CE, Lee SW, and Ree M. *J. Adh. Sci. Tech.*
27 2000;14:1131-1143.
- 28 18. Flanders DC. *Appl. Phys. Lett.* 1983;42:492-494.
- 29 19. Ibn-Elhaj M and Schadt M. *Nature* 2001;410:796-799.
- 30 20. Lloyd DR, Kinzer KE, and Tseng HS. *J. Membr. Sci.* 1990;52:239-261.
- 31 21. Lloyd DR, Kim SS, and Kinzer KE. *J. Membr. Sci.* 1991;64:1-11.
- 32 22. Matsuyama H, Ohga K, Maki T, Tearamoto M, and Nakatsuka S. *J. Appl. Polym. Sci.*
33 2003;89:3951-3955.
- 34 23. Yamaguchi M, Okada K, Abd Manaf ME, Shiroyama Y, Iwasaki T, and Okamoto K.
35 *Macromolecules* 2009;42:9034-9040.
- 36 24. Abd Manaf ME, Tsuji M, Nobukawa S, and Yamaguchi M. *Polymers* 2011;3:955-966.
- 37 25. Yamaguchi M, Iwasaki T, Okada K, and Okamoto K. *Acta Mater.* 2009;57:823-829.

- 1 26. Debye P and Bueche AM. *J. Appl. Phys.* 1949;20:518-525.
- 2 27. Pogodina NV, Siddiquee SK, van Egmond JW, and Winter HH. *Macromolecules*
3 1999;32:1167-1174.
- 4 28. Stein RS, Onogi S, Sasaguri K, and Keedy DA. *J. Appl. Phys.* 1963;34:80-89.
- 5 29. Cao SG, Shi YQ, and Chen GW. *J. Membr. Sci.* 2000;165:89-97.
- 6 30. Born M and Wolf E. Principles of optics. Chapt. 14. London: Cambridge University Press,
7 1999. pp. 735.
- 8 31. Richter I, Sun PC, Xu F, and Fainman Y. *Appl. Opt.* 1995;34:2421-2429.

9

10

1 Captions

2

3 Table 1 Glass transition and drawing temperatures (T_g and T_{draw}) for CTA and CTA/DOA films.

4

5 Table 2 Water content in stretched and immersed CTA and CTA/DOA films

6

7 Figure 1. Chemical structures of CTA and DOA.

8

9 Figure 2. Dynamic mechanical properties of unstretched CTA and CTA/DOA films.

10

11 Figure 3. Stress-strain curves for CTA and CTA/DOA films with the initial strain rate of 0.05 s^{-1} at
12 T_{draw} shown in Table 1.

13

14 Figure 4. Wavelength dispersion of birefringence Δn for CTA and CTA/DOA stretched films after
15 stretching and immersion into methanol. The increase of Δn for CTA after immersion is 2×10^{-4}
16 while that for CTA/DOA is larger than 15×10^{-4} .

17

18 Figure 5. Debye-Bueche plot for CTA/DOA film (A) before extraction of DOA and (B) after
19 extraction.

20

21 Figure 6. SEM images of CTA/DOA films after (A) solution-cast, (B) annealed, and (C) stretched.
22 The black surface represents a viewing screen for the SEM image.

23

24 Figure 7. XRD patterns of (A) CTA and (B) CTA/DOA films unstretched, stretched, and after
25 immersion.

26

27 Figure 8. Schematic illustration for the anisotropic nano-porous structure in CTA/pore system
28 prepared from the stretched CTA/DOA films.

29

30 Figure 9. Simulated result of form birefringence Δn_F in CTA/DOA and CTA/pore films. The
31 inset figures show the difference in refractive indices between CTA-DOA and CTA-pore.

32

33

1 Tables

2

3 Table 1 Glass transition and drawing temperatures (T_g and T_{draw}) for CTA and CTA/DOA films.

4

	$T_g / ^\circ\text{C}$	$T_{\text{draw}} / ^\circ\text{C}$
CTA	188	213
CTA/DOA	183	205

5

6

7

Table 2 Water content in stretched and immersed CTA and CTA/DOA films

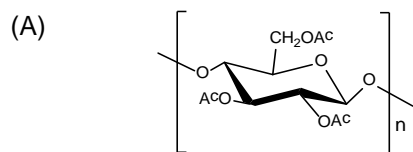
	<i>water content / wt%</i>	
	<i>stretched</i>	<i>immersed</i>
CTA	1.76	1.51
CTA/DOA	1.30	1.31

1
2
3
4
5

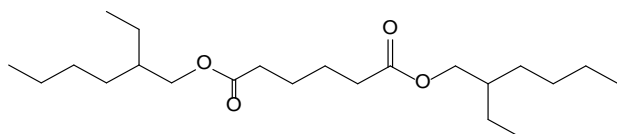
1 Figures

2

3



(B)



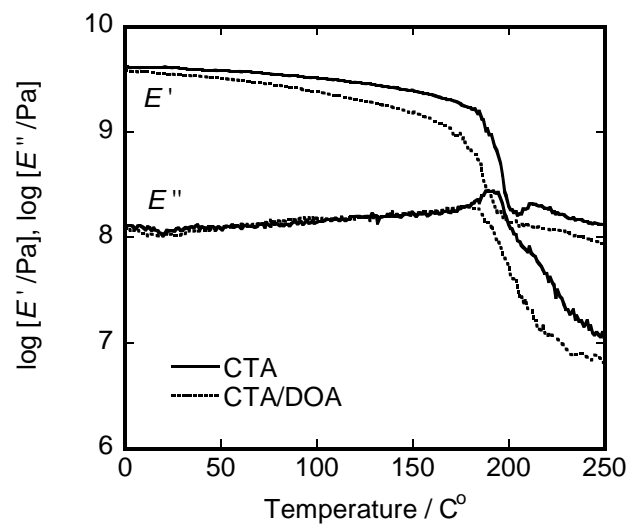
4

5

6 Figure 1. Chemical structures of CTA and DOA.

7

8

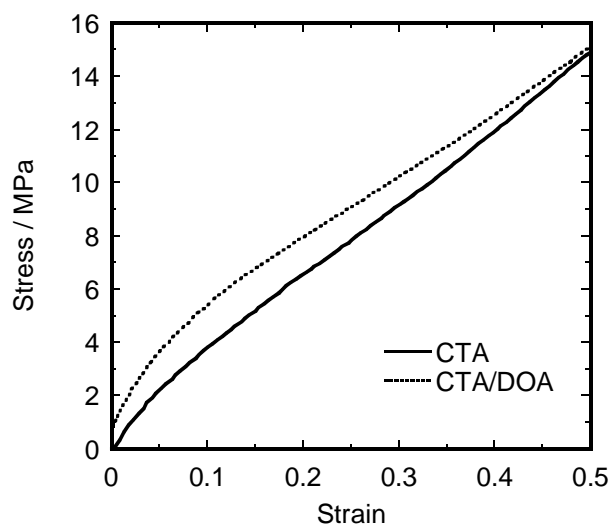


1

2

3 Figure 2. Dynamic mechanical properties of unstretched CTA and CTA/DOA films.

4

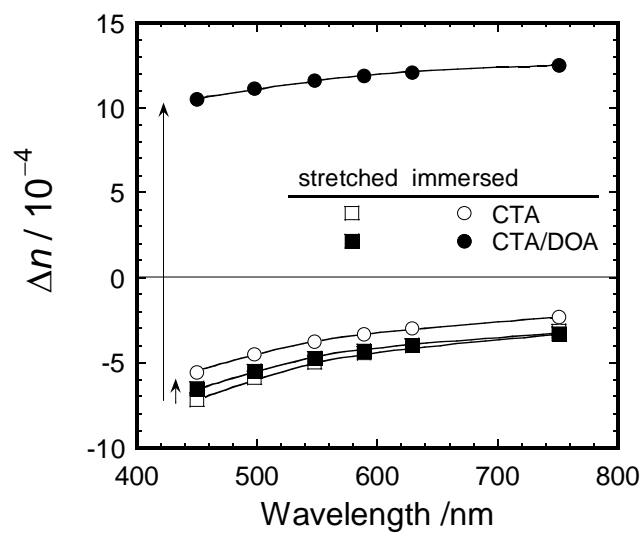


1

2

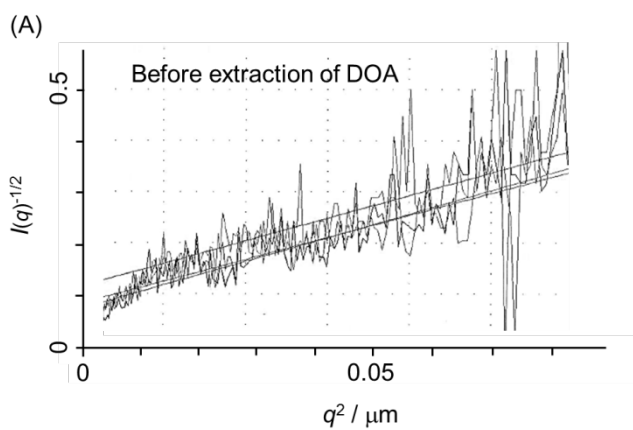
3 Figure 3. Stress-strain curves for CTA and CTA/DOA films with the initial strain rate of 0.05 s^{-1} at
4 T_{draw} shown in Table 1.

5

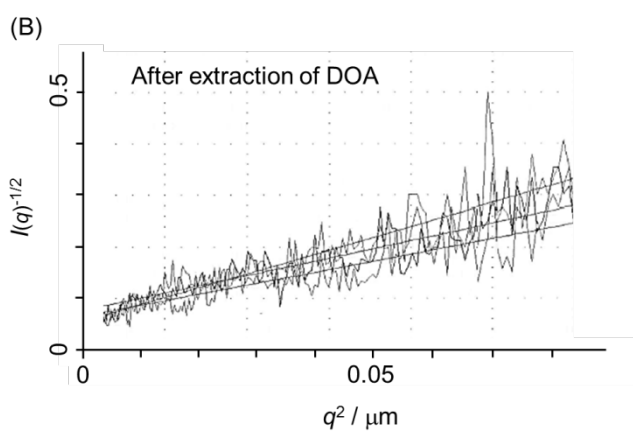


1
2
3
4
5
6
7
8

Figure 4. Wavelength dispersion of birefringence Δn for CTA and CTA/DOA films after stretching and immersion into methanol. The increase of Δn for CTA after immersion is 2×10^{-4} while that for CTA/DOA is larger than 15×10^{-4} .



1



2

3 Figure 5. Debye-Bueche plot for CTA/DOA film (A) before extraction of DOA and (B) after
4 extraction.

5

6

7

8

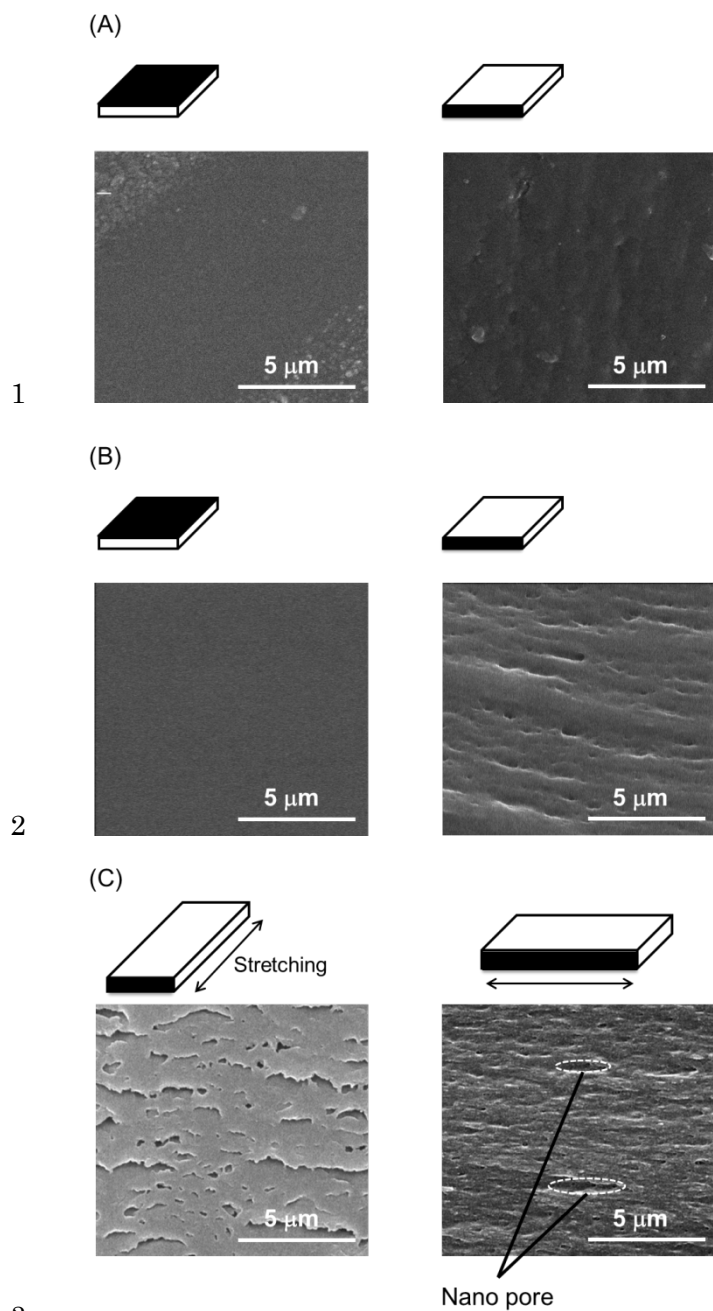
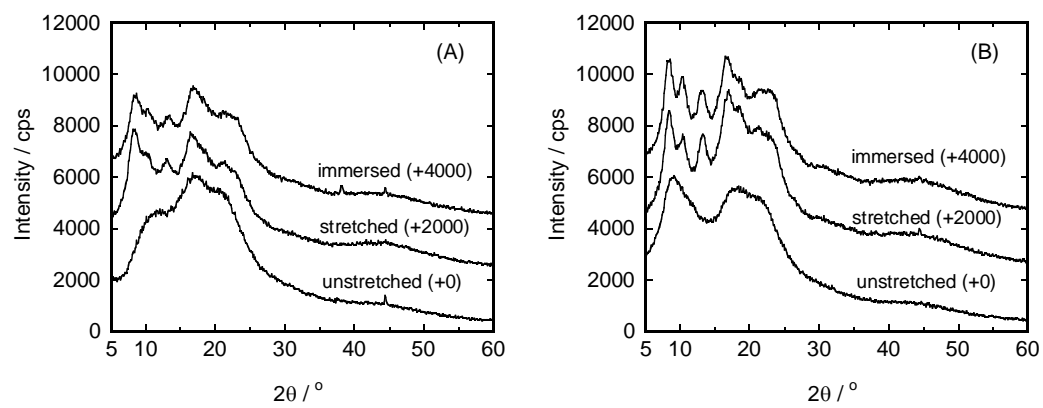


Figure 6. SEM images of CTA/DOA films after (A) solution-cast, (B) annealed, and (C) stretched. The black surface represents a viewing screen for the SEM image.



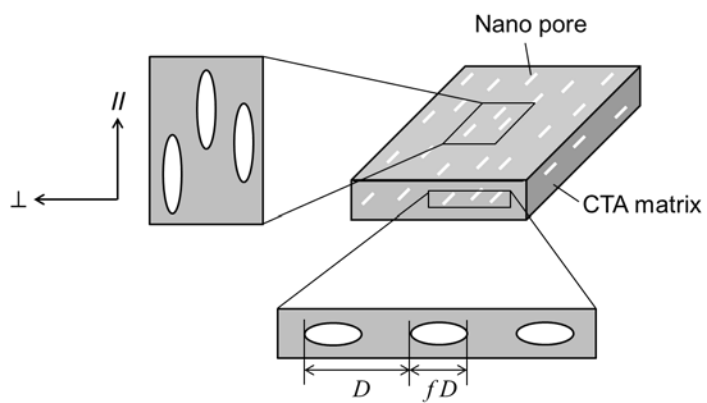
1

2

3 Figure 7. XRD patterns of (A) CTA and (B) CTA/DOA films unstretched, stretched, and after
4 immersion.

5

6



1

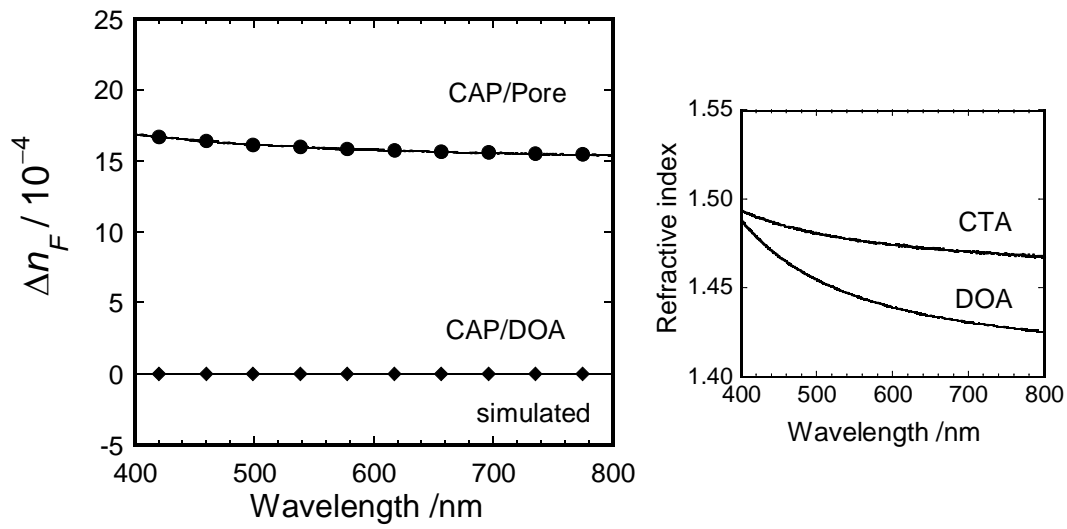
2

3 Figure 8. Schematic illustration for the anisotropic nano-porous structure in CTA/pore system
4 prepared from the stretched CTA/DOA films.

5

6

1



2

3

4 Figure 9. Simulated result of form birefringence Δn_F in CTA/DOA and CTA/pore films. The
5 inset figure shows refractive indices of CTA and DOA at various wavelengths.

6

7

8

Supporting information

These figures show NMR spectra of DOA, CTA and CTA/pore after extraction of DOA. Since the spectrum of CTA/pore indicates only CTA, DOA was completely removed by the extraction method.

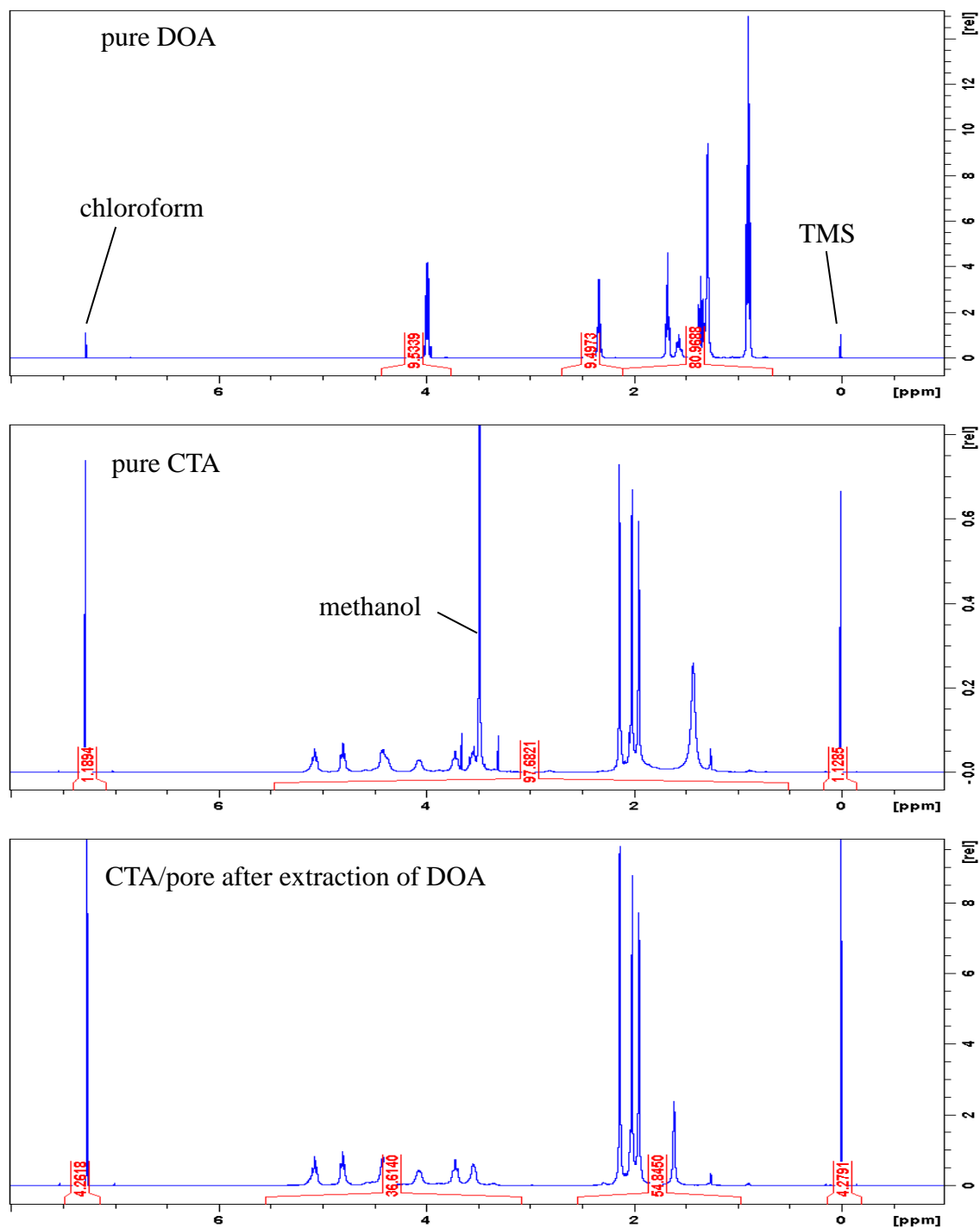


Figure S1. ¹H-NMR spectra of DOA, CTA and CTA/pore after extraction of DOA.