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Description	

Improvement of Rigidity for Rubber-Toughened Polypropylene via Localization of Carbon Nanotubes

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ABSTRACT

A methodology for controlling the localization of multi-walled carbon nanotubes (MWCNTs) was studied using an immiscible polymer blend composed of isotactic polypropylene (PP) and ethylene-co-propylene rubber (EPR). We found that the MWCNTs were localized in EPR domains within the composite, which was prepared at 280 °C. EPR molecules bonded to the MWCNT surface are responsible for the localization in EPR. Conversely, MWCNTs preferentially resided in the matrix when the composite was prepared at 190 °C along with nitrogen gas purging. Because of the selective localization of the MWCNTs in the matrix, the composite obtained via mixing at the lower temperature exhibited a higher Young's modulus and yield strength. These findings demonstrate that mixing conditions greatly affect the MWCNT distribution and thus the mechanical properties of PP/EPR blends.

Keywords: Carbon nanotubes; Polymer-matrix composites; Mechanical properties; Mixing

1 **1. Introduction**

2 Carbon nanotubes (CNTs) have received much attention in both academia and
3 industry and are recognized as being suitable candidates for use as fillers for thermoplastic
4 and thermosetting resins to enhance their Young's modulus and strength effectively; thus,
5 their range of potential applications can be significantly expanded [1–3]. In the case of
6 immiscible polymer blends with phase-separated structures, the selective localization of
7 CNTs in one of polymer phases or at the interface is crucial to achieve a material with
8 tailored properties [4–6]. For example, the rigidity and electrical conductivity can be
9 efficiently enhanced when CNTs are selectively dispersed in a continuous phase of a blend
10 with a sea-island structure. The migration of fillers to a favored polymer phase during mixing
11 usually occurs when fillers are pre-dispersed in a thermodynamically unfavorable polymer
12 [4,6,7]. Moreover, processing conditions and/or mixing protocols can have a significant
13 impact on the final morphology [4,8,9]. Furthermore, it was reported that the addition of
14 multi-walled carbon nanotubes (MWCNTs) can change the domain sizes of a phase separated
15 polymer blend, leading to the modification of its electrical and mechanical properties [9–11].

16 Isotactic polypropylene (PP) with an impact modifier such as ethylene-co-propylene
17 rubber (EPR) is widely employed in industry because of its excellent balance of both rigidity
18 and toughness [12–15]. Recently, however, the demand to enhance polymer rigidity is
19 increasing, especially for automobile applications, so that the thickness and weight of the
20 polymer components can be reduced. Hence, the focus on MWCNT additives is based on
21 wanting to improve a polymer's rigidity. For the above mentioned polymer composites, fillers
22 have to be localized in a continuous phase to effectively increase the rigidity without loss of
23 impact strength [16]. It has been reported that carbon nanofillers, such as carbon black (CB),
24 vapor-grown carbon fibers (VGCFs) and CNTs, tend to reside in polyethylene (PE) and
25 ethylene copolymers in an immiscible blend, although the interfacial tension between PE and

1 carbon nanofillers is high. Sumita et al. found that CB preferentially localized in the PE phase
2 in blends of PE and PMMA [17,18]. Wu et al. also reported that VGCFs were dispersed in PE
3 in PE/PMMA blends. They proposed the concept of an "entropy penalty" that causes a
4 selective adsorption of PE chains on the rough ends of VGCF owing to the flexibility of
5 polymer chains, which leads to a self-assembled conductive network within the PMMA
6 matrix. Even in a blend with another polyolefin, such as PP, CB was found to be selectively
7 located in the blend's PE phase [19–22]. Haddadi-Asl studied the structure and properties of
8 PP/EPR blends containing both CB and graphite fibers. He found that most conductive
9 carbon fillers resided in the EPR phase, resulting in a low proportion of fillers in the PP phase
10 [23]. Hemmati et al. studied the effect of the concentration of both single-walled carbon
11 nanotubes (SWCNTs) and their compatibilizer in PP, i.e., PP-graft-maleic anhydride (PP-g-
12 MA), on the morphology and mechanical properties of PP/ethylene-propylene-diene
13 terpolymer (EPDM) blends. They reported that the tensile modulus was greatly enhanced by
14 increasing the SWCNT content; this enhancement was more pronounced in the
15 compatibilized composite. Furthermore, EPDM was uniformly distributed in the PP matrix
16 with low concentrations of the SWCNTs and compatibilizer. However, the distribution of
17 EPDM was not homogeneous with an increased filler concentration because the
18 concentration of PP-g-MA in the matrix was also increased, thereby leading to a reduction of
19 matrix viscosity. As a result, the viscosity ratio of EPDM to PP was increased and thus the
20 break-up process of the EPDM droplets became more difficult [24].

21 The localization of CNTs in the matrix is also a key technology for realizing electro-
22 conductive composites. Meincke et al. showed that the selective localization of CNTs in
23 polyamide-6 in a blend of polyamide-6 and acrylonitrile-butadiene-styrene terpolymer greatly
24 reduced the concentration of CNTs required to form a conductive network [25]. Phromdee et
25 al. found that the volume resistivity of a PP/EPR blend with CBs was low compared with that

1 of a CB-filled PP homopolymer because of double percolation, which involves CBs creating
2 a conductive path in the continuous sections of the EPR phase [26].

3 In our previous papers, we revealed that an interphase transfer of the MWCNTs from
4 PP or PE to polycarbonate (PC) occurs during the annealing of laminated sheets composed of
5 PP/MWCNT or PE/MWCNT and PC. However, such a transfer was not detected at all from
6 PC to PP or PE. This is reasonable because the interfacial tension between PC and MWCNTs
7 is lower than that between polyolefin and MWCNTs. In contrast, MWCNTs prefer to move
8 from PC to PE during mixing at high temperature. It was revealed that this phenomenon
9 occurs owing to the adsorption of PE molecules on the surface of the MWCNTs. Furthermore,
10 the adsorption of PE is pronounced owing to the presence of oxygen during mixing at high
11 temperature [6,27]. The reaction between oxygen and the ethylene unit, which produces
12 polymer radicals, is responsible for the surface adsorption; thus, a similar phenomenon is
13 expected to occur for copolymers of ethylene and α -olefin such as EPR. Based on this
14 principle, we improved the mixing technique to control the distribution of MWCNTs in
15 immiscible blends with a sea-island structure.

16 In this research, a blend of PP and EPR was employed as it is one of the most
17 important immiscible blends used in industrial applications. We controlled the MWCNT
18 distribution via the processing/mixing conditions, e.g., via the temperature and the nitrogen
19 gas purging conditions. Our experimental results provide a method to effectively enhance the
20 rigidity via the selective localization of MWCNTs in the matrix phase.

21

22 **2. Experimental Section**

23 *2.1 Materials*

24 The polymers used in this research were commercially available isotactic
25 polypropylene PP (J108M, Prime Polymer, Japan) and ethylene-co-propylene rubber EPR

1 (EP11, JSR, Japan). The stereoregularity of PP is 96% (*mmmm*%, a meso-pentad value). The
2 melt flow rate of PP is 45 g/10min at 230 °C. The number- and weight-average molecular
3 weights of PP, determined by size exclusion chromatography, are 3.6×10^4 and 2.5×10^5 ,
4 respectively, versus a polypropylene standard. The ethylene content in EPR is 52 wt.%, and
5 the Mooney viscosity $ML_{(1+4)} 100\text{ °C}$ is 40.

6 A composite with 20 wt.% multi-walled carbon nanotubes, i.e., PP/MWCNT, was
7 kindly provided by Hodogaya Chemical Co., Ltd., Japan in pellet form. The MWCNTs (NT-7,
8 Hodogaya Chemical, Japan) were produced by a catalytic chemical vapor deposition
9 technique using a floating reactant method and subsequent thermal treatment up to 2600 °C
10 [28,29]. The diameter of the MWCNTs ranged from 40 to 80 nm, and the length was between
11 10 and 20 μm with a density of 2300 kg/m^3 .

12

13 2.2 Sample preparation

14 PP/MWCNT (80/20) was mixed with neat EPR at either 190 or 280 °C for 10 min
15 with a blade rotation speed of 50 rpm with a weight ratio of 80/20, i.e., PP/MWCNT/EPR =
16 64/16/20. The temperatures were selected as the lowest and highest ones for conventional
17 processing operations of PP. In order to prevent marked chain scission reaction, 5000 ppm of
18 a thermal stabilizer, 2-(tert-butyl)-6-methyl-4-(3-((2,4,8,10-tetrakis(tert-butyl)dibenzo[d,f]
19 [1,3,2]dioxaphosphin-6-yl)oxy)propyl)phenol (Sumilizer-GP, Sumitomo Chemical, Japan)
20 was added to all samples. Furthermore, the effect of nitrogen N_2 gas purging on the structure
21 was investigated. Additionally, a blend of PP/EPR with a weight ratio of 76/24 (= 64/20) was
22 also prepared at 190 °C for 10 min with 5000 ppm of the thermal stabilizer (no N_2 gas).

23 All samples were compressed into flat sheets with a thickness of 1 mm using a
24 laboratory compression-molding machine (Table-type test-press, Tester Sangyo, Japan) at
25 190 °C under 10 MPa for 3 min and subsequently cooled at 25 °C for 3 min.

1

2 *2.3 Measurements*

3 The quenched PP/MWCNT/EPR sheets were crushed into powder form using a
4 cryogenic sample crusher (JFC-300, Japan Analytical Industry, Japan). The powder was
5 packed in a metal mesh bag and then immersed in xylene at room temperature for three days.
6 Thereafter, the insoluble part, which is the part remaining after the EPR extraction, was taken
7 out and dried to quantify its weight. The collected solution was also dried to obtain the
8 soluble part. The weight fraction of the soluble part S was calculated using Eq. (1):

$$9 \quad S = \frac{w_i - w_f}{w_i} \times 100, \quad (1)$$

10 where w_i and w_f are the weights of the initial sample and dried insoluble one, respectively.

11 The molecular weight of the soluble part in 1,2,4-trichlorobenzene at 140 °C was
12 measured by size exclusion chromatography (HLC-8321GPC/HT, Tosoh, Japan).

13 Rheological measurements for PP, EPR, PP/EPR and PP/MWCNT/EPR were carried
14 out using a parallel-plate rheometer (AR 2000ex, TA Instruments, Japan) at 170 and 190 °C
15 under an N₂ atmosphere. The diameter of the plates was 8 mm for the composites and 25 mm
16 for the samples without MWCNTs. All samples were kept in the rheometer for 10 min prior
17 to the measurements.

18 The morphology and localization of the MWCNTs in the PP/MWCNT/EPR blends
19 were investigated using scanning electron microscopy (SEM) (S4100, Hitachi, Japan) and
20 transmission electron microscopy (TEM) (JEM-2000FX, JEOL, Japan). For SEM
21 observations, the sheets were cryofractured in liquid N₂ and etched with xylene at room
22 temperature for 3 days to remove the EPR fraction on the surface of the blend. The etched
23 samples were coated with Pt-Pd by a sputter coating machine prior to the observations. For

1 TEM, the sheets were prepared by cutting a cryosection using an ultra-microtome (MT-XL,
2 RMC-Boeckeler, AZ, USA) with a diamond knife after exposure to ruthenium tetroxide.

3 Electrical resistivity measurements were carried out on the surface of the compressed
4 films using a constant-current supply resistivity meter (MCP-T610, Mitsubishi Chemical
5 Analytech, Japan). The resistivity was measured five times for each sample at room
6 temperature and the average value was calculated.

7 The temperature dependence of the dynamic tensile moduli was measured using a
8 dynamic mechanical analyzer (Rheogel-E4000, UBM, Japan) at a heating rate of 2 °C/min
9 over a temperature range from -80 to 160 °C. The applied frequency was 10 Hz.

10 Tensile tests were performed at room temperature using a universal tensile testing
11 instrument (LSC-50/300, Tokyo Testing Machine, Japan) at a constant crosshead speed of 50
12 mm/min (strain rate of 0.1 min⁻¹) according to the ASTM D638 standard test method. The
13 specimens were dumbbell shaped (ASTM D638-V) films with a thickness of 0.5 mm and had
14 a gauge length of 10 mm.

15

16 **3. Results and discussion**

17 Fig. 1 shows the shear storage modulus G' and loss modulus G'' as a function of
18 angular frequency ω for individual pure polymers and their blends at 190 °C. It was found
19 that both moduli of EPR were significantly higher than those of PP because of EPR's longer
20 relaxation time caused by its high molecular weight. The zero-shear viscosities of the
21 polymers at 190 °C were 730 Pa s for PP and 45,000 Pa s for EPR. In the case of the PP/EPR
22 blend, a long relaxation mechanism ascribed to the phase separation is clearly visible in the
23 $G'(\omega)$ curve. This is a typical phenomenon observed for an immiscible polymer blend with
24 high viscous dispersions in a low viscous matrix [30–32].

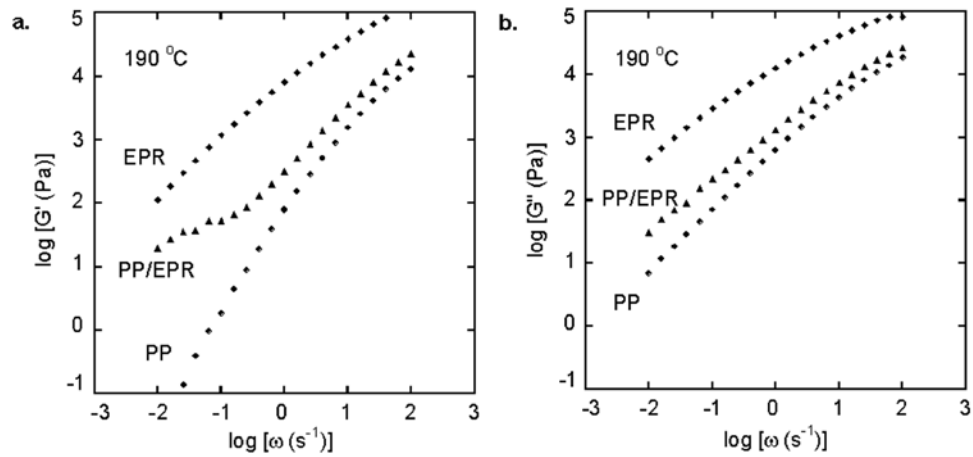


Fig. 1. Frequency dependence of oscillatory shear moduli such as (a) storage modulus G' and (b) loss modulus G'' for PP, EPR and PP/EPR (64/20) at 190 °C.

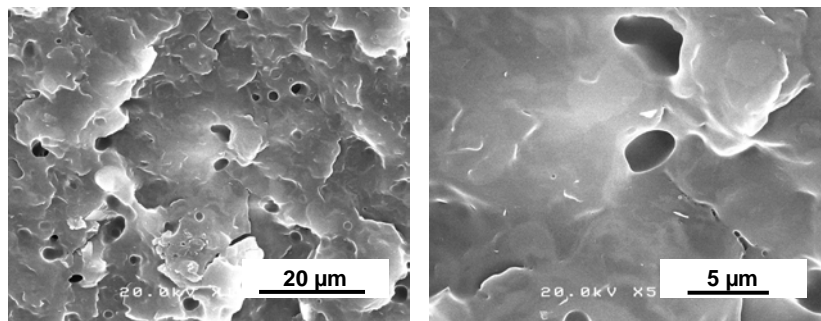
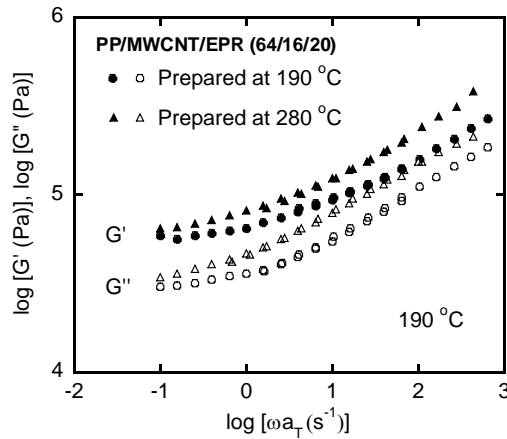


Fig. 2. SEM image of fractured surface of PP/EPR (64/20) after removal of EPR.

The morphology of the PP/EPR (64/20) blend was observed by SEM using the cryofractured surface as shown in Fig. 2. The EPR fraction was removed by xylene at room temperature. Prior to the experiment, it was confirmed that a pure EPR sheet is dissolved in xylene under the same experimental conditions. Fig. 2 shows the SEM image of the PP/EPR (64/20) blend. A sea-island structure in which EPR domains are dispersed in the PP matrix is clearly visible. It can be observed that the dispersed EPR domains are large and irregular in the shape owing to the long relaxation time of EPR.

1 PP/MWCNT and EPR were mixed at various conditions to establish a mixing
 2 technique to control the MWCNT distribution and thus the mechanical properties of the blend
 3 with MWCNTs. Besides, the effect of N₂ purging was also examined.



4
 5 **Fig. 3.** Master curves of frequency dependence of oscillatory shear moduli for PP/MWCNT/
 6 EPR composites prepared at 190 °C (circles) and 280 °C (triangle) without N₂ purging.

7
 8 In Fig. 3, we demonstrated that both G' and G'' of the composite prepared by mixing
 9 at 190 °C were higher than those of the composite prepared at 280 °C. Both samples were
 10 prepared without N₂ purging. The results suggest that the mixing temperature greatly affects
 11 the rheological responses, which is ascribed to the difference in the MWCNT distribution.
 12 Since the moduli of the sample prepared at the lower temperature are higher than those of the
 13 one prepared at the higher temperature, more MWCNTs could be distributed in the
 14 continuous phase, i.e., PP, when mixed at the lower temperature. We also found that the
 15 change in the shear moduli caused by the introduction of N₂ was barely detectable (data not
 16 shown here).

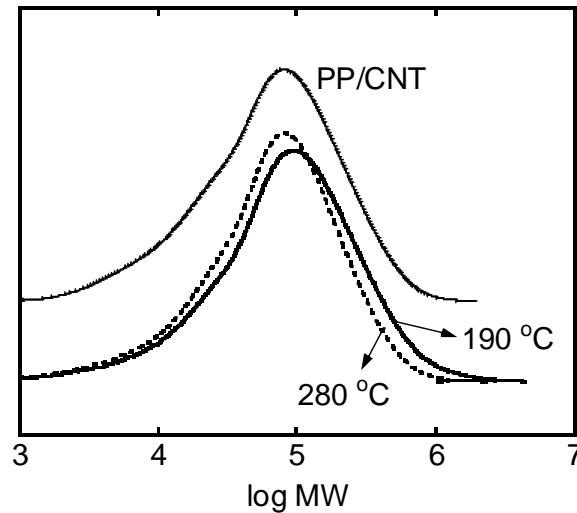
17 Although the rheological responses were similar irrespective of the N₂ purging, the
 18 surface resistivity was changed as follows; 12.2 Ω/sq. for the sample prepared at 280 °C
 19 without N₂ purging, 10.0 Ω/sq. for the sample prepared at 190 °C without N₂ purging, and

1 6.5 Ω /sq. for the sample prepared at 190 °C with N₂ purging, respectively. This difference is
2 attributed to the MWCNT localization in the composite; when more MWCNTs reside in the
3 PP phase, the conductive path builds up more easily in the matrix, leading to a low electrical
4 resistivity. The present results suggest that the surface resistivity is more sensitive to the
5 uneven distribution of the MWCNTs. Moreover, it should be noted that the N₂ purging
6 enhances the MWCNT dispersion in the matrix to some degree.

7 Furthermore, the weight fraction of the insoluble part in xylene at room temperature
8 was measured using the sample powder. It was found that the weight fraction of the
9 composite prepared at the lower temperature with N₂ purging was 79.5 wt.%, which
10 corresponds to the weight proportion of PP and MWCNTs in the composite. This
11 demonstrates that most MWCNTs were located in the PP phase. When the sample was
12 prepared without N₂ purging at 190 °C, it was 82.0 wt.%. For the composite prepared at
13 280 °C without N₂ purging, the weight fraction of the insoluble part was 87.7 wt.%,
14 suggesting that the EPR was not dissolved completely in the solvent. This result indicates that
15 undissolved EPR (7.7 wt.%) was adsorbed on the MWCNT surface.

16 The molecular weight of PP in the composites was measured using the soluble part in
17 1,2,4-trichlorobenzene at 140 °C. As shown in Fig. 4, the molecular weight and its
18 distribution of the composite prepared at 280 °C without N₂ purging were almost the same
19 with those of the PP/MWCNT, although the composite mixed at 190 °C with N₂ purging
20 shows higher molecular weight due to the dissolution of EPR fraction. The result
21 demonstrates that chain scission reaction of PP does not take place drastically even at 280 °C,
22 presumably owing to a large amount of the thermal stabilizer.

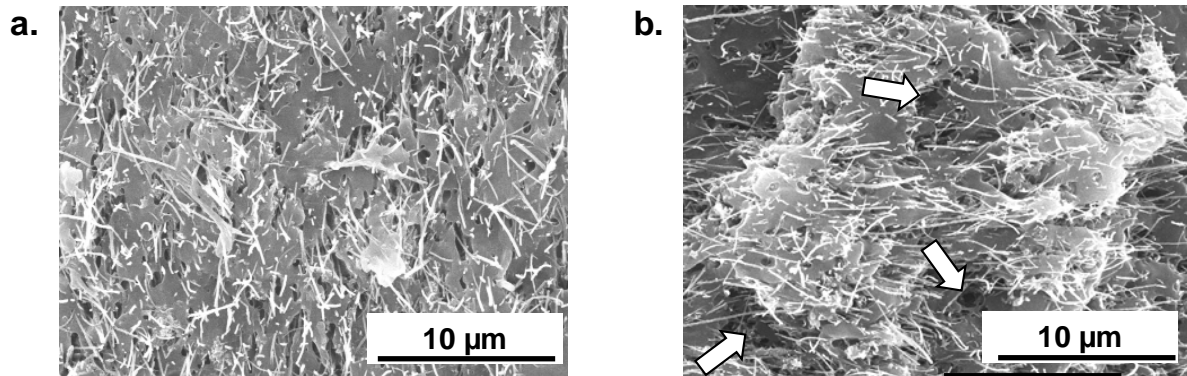
23 To identify the MWCNT dispersion in the composite, SEM observations were
24 performed using the composites prepared with different mixing conditions (Fig. 5).



1

2 **Fig. 4.** Molecular weight distribution of soluble parts in 1,2,4-trichlorobenzene at 140 °C for
 3 PP/MWCNT (thin solid line) and PP/MWCNT/EPR composites prepared at 190 °C with N₂
 4 purging (bold solid line) and 280 °C without N₂ purging (bold dotted line).

5



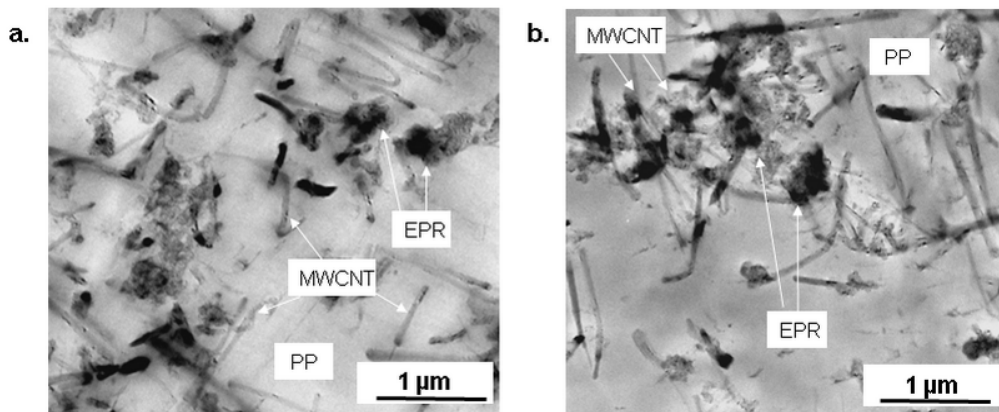
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7 **Fig. 5.** SEM images of fractured surfaces of PP/MWCNT/EPR composites prepared at (a)
 8 190 °C with N₂ purging and (b) 280 °C without N₂ purging. EPR fraction was removed by
 9 xylene prior to Pt-Pd coating.

10

11 Because the fractured sample was immersed in xylene, we attribute the holes detected
 12 in both samples to the dispersion pattern of EPR. Compared with the blend without
 13 MWCNTs (Fig. 2), the domain sizes of EPR were obviously small after the incorporation of
 14 the MWCNTs. Considering that MWCNTs were added in PP first, the fine dispersion of EPR

1 is attributed to the change in the viscosity ratio between PP and EPR. A fine dispersion is
 2 known to be obtained when the viscosity ratio is close to 1 [33,34]. In the composite prepared
 3 at 280 °C without N₂ purging, some large holes are detected, which are denoted by arrows,
 4 compared with that prepared at 190 °C with N₂ purging. Once the MWCNTs moved to EPR,
 5 the viscosity ratio increased significantly and therefore the morphology became coarse. The
 6 effect of the N₂ purging on the morphology was not detected by the SEM observation.



7

8 **Fig. 6.** TEM images of PP/MWCNT/EPR composites prepared at (a) 190 °C with N₂ purging
 9 and (b) 280 °C without N₂ purging.

10

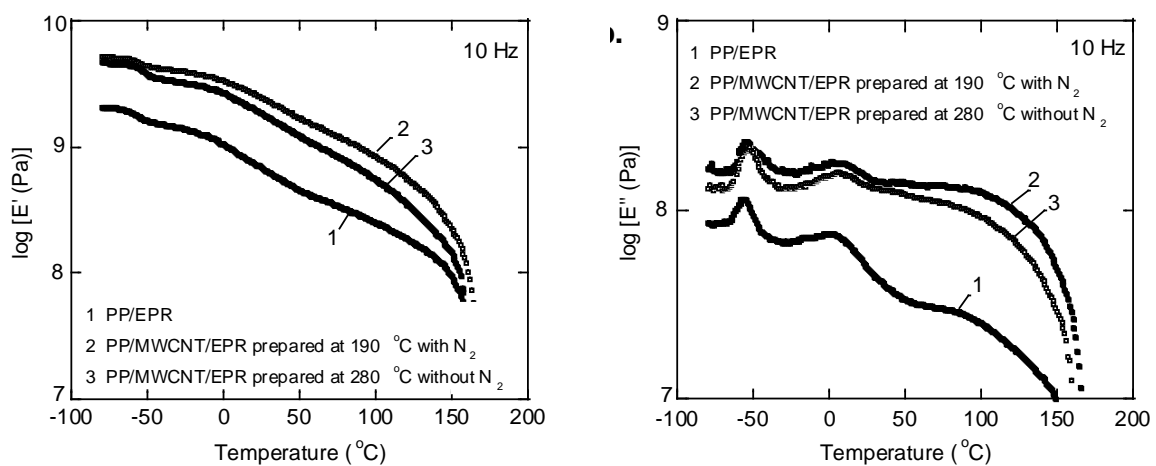
11 The MWCNT dispersion was also examined via TEM. Fig. 6 shows TEM images of
 12 the PP/MWCNT/EPR composites. The grey continuous phase is PP, while the dark dispersed
 13 phase is EPR. As seen in the figure, more MWCNTs were localized in EPR when the sample
 14 was prepared at 280 °C without N₂ purging, which is in agreement with the conclusions
 15 drawn from the SEM images and solvent immersion experiments.

16

17 These results indicate that a large amount of MWCNTs are localized in the PP phase
 18 when PP/MWCNT/EPR was prepared at 190 °C with the introduction of N₂ gas. In contrast,
 19 the MWCNTs were transferred from PP into the EPR phase at the higher mixing temperature
 20 and distributed within it. The MWCNT migration occurs owing to the adsorption of EPR on
 the MWNCT surface at the high temperature, presumably with the aid of oxygen, which

1 promotes the generation of free radicals of the ethylene unit in EPR molecular chains
2 [6,35,36].

3 Mechanical properties were evaluated to clarify the effect of the MWCNT
4 localization. Temperature dependence of the tensile storage modulus E' and loss modulus E''
5 for the neat PP/EPR and the PP/MWCNT/EPR composites that have different structures is
6 shown in Fig. 7. The glass transition temperature T_g of each sample was obtained from the
7 peak temperature in the E'' curves and the results are listed in Table 1.



8
9 **Fig. 7.** Temperature dependence of (a) tensile storage modulus E' and (b) loss modulus E'' for
10 PP/EPR (1) and PP/MWCNT/EPR prepared by different mixing conditions (190 °C with N_2
11 purging (2) and 280 °C without N_2 purging (3)).

13 **Table 1.** Glass transition temperature of PP/EPR and PP/MWCNT/EPR

Samples	T_g (°C)	
	EPR	PP
PP/EPR	-56.1	4.8
PP/MWCNT/EPR prepared at 190 °C with N_2	-55.3	5.3
PP/MWCNT/EPR prepared at 280 °C without N_2	-52.3	5.4

14

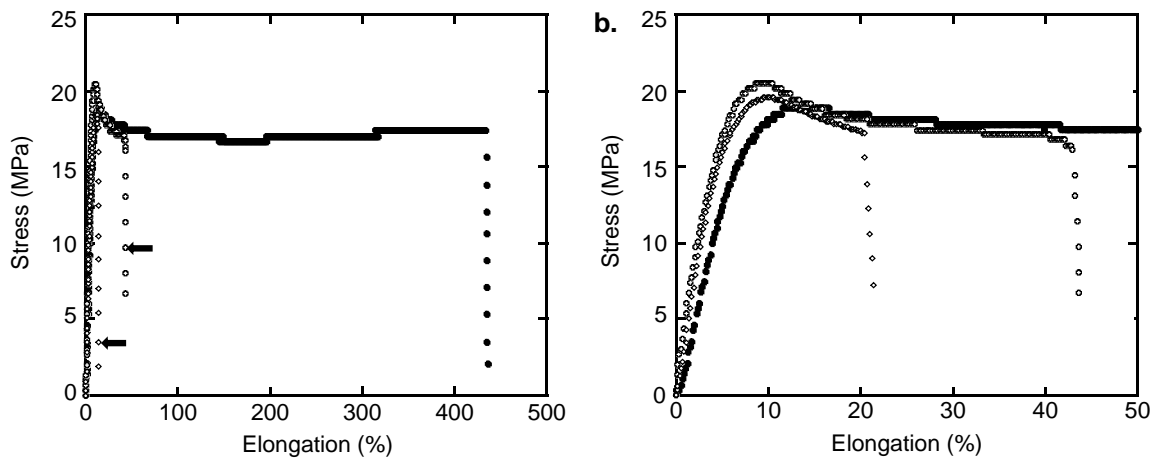
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1 We found that E' is greatly enhanced by the addition of MWCNTs owing to the filling
2 effect of rigid materials [1,30,39], which was more pronounced for the composite prepared at
3 190 °C with N₂ purging. This is reasonable since more MWCNTs reside in the matrix.
4 Moreover, all the samples show double peaks in the E'' curve, corresponding to the T_g values
5 of the EPR and PP phases. As seen in Fig. 7(b), the T_g s of EPR and PP in the binary blend are
6 -56.1 and 4.8 °C, respectively. We did the same measurements three times for each sample
7 and confirmed that the difference among the samples is reproducible. Furthermore, the T_g of
8 PP is shifted slightly to a higher temperature by the addition of the MWCNTs with an
9 enhanced α relaxation mode. The well-developed crystalline structure, presumably caused by
10 the nucleating ability of the MWCNTs [24,37,38], is responsible for both the prominent α
11 relaxation mode as well as the T_g shift of PP (owing to the restricted motion of amorphous
12 chains by crystals). Moreover, it should be noted that the T_g of EPR in the composite
13 prepared at 280 °C without N₂ purging is located at a higher temperature than that of EPR in
14 the PP/EPR blend without the MWCNTs. This is attributed to the reduced motion of the EPR
15 molecules adsorbed on the surfaces of the MWCNTs. Finally, it should be noted that the
16 storage modulus of the composite prepared at 190 °C with N₂ purging is greatly higher than
17 that at 280 °C without N₂ purging in the wide temperature region. This result demonstrates
18 that the rigidity is enhanced efficiently by the MWCNT localization in the matrix.

19 Fig. 8 shows the stress–strain curves. The tensile properties are summarized in Table
20 2. It is reasonably found that the PP/MWCNT/EPR composites show higher yield stress than
21 the PP/EPR blend owing to the reinforcing ability of the MWCNTs, which is pronounced for
22 the composite prepared at 190 °C with N₂ purging. This is plausible because the MWCNTs,
23 which are dispersed in the matrix, enhance the stiffness with a consequent increase in the
24 modulus. Further, it should be noted that the material shows a large elongation at break,
25 demonstrating that the fracture energy is significantly larger than that of the composite

1 prepared at 280 °C without N₂ purging. When MWCNTs are located in the EPR phase,
 2 leading to the modulus enhancement of rubber dispersions, the reinforcing capability was
 3 greatly reduced. The result indicates that the enhancement of the tensile properties is
 4 dependent on the mixing condition.

5



6

7 **Fig. 8.** Stress-strain curves for PP/EPR and PP/MWCNT/EPR prepared by different mixing
 8 conditions (190 °C with N₂ purging and 280 °C without N₂ purging). The initial parts in the
 9 small strain region are shown in (b).

10

11

Table 2. Tensile properties of PP/EPR and PP/MWCNT/EPR

Samples	Yield stress (MPa)	Elongation at break (%)
PP/EPR	18.6±0.3	487±65
PP/MWCNT/EPR prepared at 190 °C with N ₂	20.6±0.8	47±10
PP/MWCNT/EPR prepared at 280 °C without N ₂	19.5±0.3	20±5

12

13

14

15

16

17

Overall, the findings demonstrate that the localization of MWCNTs can be controlled mostly via the mixing temperature as presented in the PP/MWCNT/EPR composites. Moreover, mixing at 190 °C with the addition of N₂ purging was effective at controlling the preferential localization of the MWCNTs in the PP matrix of the PP/EPR blend with the sea-

1 island structure. This technique is available for other polymer blends, at least, those
2 containing EPR as an impact modifier, because the adsorption of EPR molecules on the
3 surface of the MWCNTs is responsible for making the localization of the MWCNTs in the
4 rubbery phase. We also expect that this concept will widen material design possibilities to
5 develop an advanced rubber-toughened plastic with high rigidity and high impact strength.

6

7 **4. Conclusions**

8 The effect of the mixing temperature along with the introduction of N₂ gas on the
9 localization of MWCNTs in immiscible PP/EPR blends was studied. The results demonstrate
10 that the MWCNTs were dispersed in the continuous PP phase when the composite was mixed
11 at low temperature. Furthermore, N₂ purging was also effective to the preferential dispersion
12 of MWCNTs in the matrix to some degree. When the mixing temperature was increased up to
13 280 °C without the use of N₂ gas, more MWCNTs were distributed in the dispersed EPR
14 phase, which we attribute to the adsorption of EPR molecules on the MWCNT surface during
15 melt mixing. As expected, when the MWCNTs were localized in the PP matrix, both the
16 Young's modulus and yield stress were higher than those for the sample in which the
17 MWCNTs were confined to the EPR phase. Moreover, the elongation at break was larger
18 than that for the composite mixed at 280 °C without N₂ purging. The results prove that the
19 rigidity of PP/EPR blends was improved by appropriate mixing conditions, i.e., at low
20 temperature with the introduction of N₂ gas, because MWCNTs were selectively localized in
21 the matrix.

22

23 **Acknowledgement**

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3

4 **References**

5 [1] S. Iijima, Helical microtubules of graphitic carbon, *Nature* 354 (1991) 56-58.

6 [2] E. Logakis, C. Pandis, V. Peoglos, P. Pissis, C. Stergiou, J. Pionteck, P. Pötschke, M.
7 Mičušík, M. Omastová, Structure–property relationships in polyamide 6/multi-walled
8 carbon nanotube nanocomposites, *J. Polym. Sci. Part B Polym. Phys.* 47, (2009) 764-774.

9 [3] M.H. Al-Saleh, U. Sundararaj, Microstructure, electrical, and electromagnetic
10 interference shielding properties of carbon nanotube/acrylonitrile–butadiene–styrene
11 nanocomposites, *J. Polym. Sci. Part B Polym. Phys.* 50 (2012) 1356-1362.

12 [4] Y.P. Mamunya, Morphology and percolation conductivity of polymer blends containing
13 carbon black, *J. Macromol. Sci. B* 38 (1999) 615-622.

14 [5] K. Prashantha, J. Soulestin, M.F. Lacrampe, P. Krawczak, G. Dupin, M. Claes,
15 Masterbatch-based multi-walled carbon nanotube filled polypropylene nanocomposites:
16 assessment of rheological and mechanical properties, *Compos. Sci. Technol.* 69 (2009)
17 1756-1763.

18 [6] R. Wiwattananukul, Y. Hachiya, T. Endo, S. Nobukawa, M. Yamaguchi, Anomalous
19 transfer phenomenon of carbon nanotube in the blend of polyethylene and polycarbonate,
20 *Compos. Part B* 78 (2015) 409-414.

21 [7] A.V. Doan, S. Nobukawa, S. Ohtsubo, T. Tada, M. Yamaguchi, Selective migration of
22 silica particles between rubbers, *J. Polym. Res.* 20 (2013) 145-150.

23 [8] A. Gödel, A. Kasaliwal, P. Pötschke, Selective localization and migration of
24 multiwalled carbon nanotubes in blends of polycarbonate and poly(styrene-acrylonitrile),
25 *Macromol. Rapid Commun.* 30 (2009) 423-429.

- 1 [9] A.C. Baudouin, J. Devaux, C. Bailly, Localization of carbon nanotubes at the interface in
2 blends of polyamide and ethylene–acrylate copolymer, *Polymer* 51 (2010) 1341-1354.
- 3 [10] P. Pötschke, S. Pegel, M. Claes, D. Bonduel, A novel strategy to incorporate carbon
4 nanotubes into thermoplastic matrices, *Macromol. Rapid Commun.* 29 (2008) 244-251.
- 5 [11] L. Zonder, A. Ophir, S. Kenig, S. McCarthy, The effect of carbon nanotubes on the
6 rheology and electrical resistivity of polyamide 12/high density polyethylene blends,
7 *Polymer* 52 (2011) 5085-5091.
- 8 [12] M. Jaziri, N. Mnif, V. Massardier-Nageotte, H. Perier-Camby, Rheological, thermal, and
9 morphological properties of blends based on polypropylene, ethylene-propylene rubber,
10 and ethylene-1-octene copolymer that could result from end of life vehicles: Effect of
11 maleic anhydride grafted polypropylene, *Polym. Eng. Sci.* 47 (2007) 1009-1115.
- 12 [13] N. Mnif, V. Massardier, T. Kallel, B. Elleuch, Effects of nanoparticle treatment and
13 compatibilizers on the properties of (PP/EPR)/nano-CaCO₃ blends, *Int. J. Mater. Form.* 1
14 (2008) 639-643.
- 15 [14] M. Yamaguchi, Y. Irie, P. Phulkerd, H. Hagihara, S. Hirayama, S. Sasaki, Plywood-like
16 structure of injection-moulded polypropylene, *Polymer* 51 (2010) 5983-5989.
- 17 [15] L. Moballegh, S. Hakim, J. Morshedian, M. Nekoomanesh, A new approach to increase
18 toughness of synthesized PP/EPR in-reactor blends by introducing a copolymerization
19 step under low ethylene concentration, *J. Polym. Res.* 22 (2015) 1-11.
- 20 [16] Y. Liu, M. Kontopoulou, The structure and properties of polypropylene and
21 thermoplastic olefin nanocomposites containing nanosilica, *Polymer* 47 (2006) 7731-
22 7739.
- 23 [17] M. Sumita, K. Sakata, S. Asai, K. Miyasaka, H. Nakagawa, Dispersion of fillers and the
24 electrical conductivity of polymer blends filled with carbon black, *Polym. Bull.* 25
25 (1991) 265-271.

- 1 [18] M. Sumita, K. Sakata, Y. Hayakawa, S. Asai, K. Miyasaka, M. Tanemura, Double
2 percolation effect on the electrical conductivity of conductive particles filled polymer
3 blends, *Colloid Polym. Sci.* 270 (1992) 134-139.
- 4 [19] R. Uotila, U. Hippi, S. Paavola, J. Seppälä, Compatibilization of
5 PP/elastomer/microsilica composites with functionalized polyolefins: Effect on
6 microstructure and mechanical properties, *Polymer* 46 (2005) 7923-7930.
- 7 [20] G. Wu, S. Asai, M. Sumita, A self-assembled electric conductive network in short carbon
8 fiber filled poly(methyl methacrylate) composites with selective adsorption of
9 polyethylene, *Macromolecules* 32 (1999) 3534-3536.
- 10 [21] G. Wu, S. Asai, M. Sumita, Entropy penalty-induced self-assembly in carbon black or
11 carbon fiber filled polymer blends, *Macromolecules* 35 (2002) 945-951.
- 12 [22] H. Yui, G. Wu, H. Sano, M. Sumita, K. Kino, Morphology and electrical conductivity of
13 injection-molded polypropylene/carbon black composites with addition of high-density
14 polyethylene, *Polymer* 47 (2006) 3599-3608.
- 15 [23] V. Haddadi-Asl, Morphology and properties of conductive carbon/polyolefins composite,
16 *Iran. Polym. J.* 5 (1996) 75-86.
- 17 [24] M. Hemmati, A. Narimani, H. Shariatpanahi, A. Fereidoon, M. Ghorbanzadeh Ahangari,
18 Study on morphology, rheology and mechanical properties of thermoplastic elastomer
19 polyolefin (TPO)/carbon nanotube nanocomposites with reference to the effect of
20 polypropylene-grafted-maleic anhydride (PP-g-MA) as a compatibilizer, *Int. J. Polym.*
21 *Mater.* 60 (2011) 384-397.
- 22 [25] O. Meincke, D. Kaempfer, H. Weickmann, C. Friedrich, M. Vathauer, H. Warth,
23 Mechanical properties and electrical conductivity of carbon-nanotube filled polyamide 6
24 and its blends with acrylonitrile-butadiene-styrene, *Polymer* 45 (2004) 739-748.

- 1 [26] S. Phromdee, P. Larjai, S. Tiptipakorn, S. Rimdusit, Development of electronic packages
2 from polypropylene/ethylene-propylene rubber blends: A double percolation approach, *J.*
3 *Met. Mater. Miner.* 24 (2014) 33-36.
- 4 [27] H. Yoon, K. Okamoto, M. Yamaguchi, Carbon nanotube imprinting on a polymer
5 surface, *Carbon* 47 (2009) 2840-2846.
- 6 [28] Y.A. Kim, T. Hayashi, M. Endo, Y. Kaburagi, T. Tsukada, J. Shan, Synthesis and
7 structural characterization of thin multi-walled carbon nanotubes with a partially faceted
8 cross section by a floating reactant method, *Carbon* 43 (2005) 2243-2250.
- 9 [29] J. Chen, J.Y. Shan, T. Tsukada, F. Munekane, A. Kuno, M. Matsuo, T. Hayashi, Y.A.
10 Kim, M. Endo, The structural evolution of thin multi-walled carbon nanotubes during
11 isothermal annealing, *Carbon* 45 (2007) 274-280.
- 12 [30] Y. Niu, L. Yang, K. Shimizu, J.A. Pathak, H. Wang, Z. Wang, Investigation on phase
13 separation kinetics of polyolefin blends through combination of viscoelasticity and
14 morphology, *J. Phys. Chem. B* 113 (2009) 8820-8827.
- 15 [31] M. Yamaguchi, T. Yokohara, B.M.A. Mohd Amran, Effect of flexible fibers on
16 rheological properties of poly(lactic acid), *Nihon Reoroji Gakkaishi*, 41 (2013) 129-135.
- 17 [32] J. Seemork, T. Sako, B.M.A. Mohd Amran, M. Yamaguchi, Rheological response under
18 non-isothermal stretching for immiscible blends of isotactic polypropylene and acrylate
19 polymer, *J. Rheology* 61 (2017) 1-11.
- 20 [33] X.Q. Liu, W. Yang, B.H. Vie, M.B. Yang, Influence of multiwall carbon nanotubes on
21 the morphology, melting, crystallization and mechanical properties of polyamide
22 6/acrylonitrile-butadiene-styrene blends, *Mater. Design* 34 (2012) 355-362.
- 23 [34] J.P. Cao, J. Zhao, F. You, H. Yu, G.H. Hu, Z.M. Dand, High thermal conductivity and
24 high electrical resistivity of poly(vinylidene fluoride)/polystyrene blends by controlling
25 the localization of hybrid fillers, *Compos. Sci. Technol.* 89 (2013) 142-148.

- 1 [35] A.J. Peacock, Handbook of polyethylene structure, properties, and applications, Marcel
2 Dekker, New York, 2000.
- 3 [36] M. Sirirumpoonthum, S. Nobukawa, Y. Satoh, H. Sasaki, M. Yamaguchi, Effect of
4 thermal modification on rheological properties of polyethylene blends, *J. Rheol.* 58
5 (2014) 449-466.
- 6 [37] L. Valentini, J. Biagiotti, M.A. Lopez-Manchado, S. Santucci, J.M. Kenny, Effects of
7 carbon nanotubes on the crystallization behavior of polypropylene, *Polym. Eng. Sci.* 44
8 (2004) 303-311.
- 9 [38] B.P. Grady, Carbon nanotube-polymer composites: manufacture, properties, and
10 applications, Wiley, Hoboken, 2011.
- 11 [39] R. Wiwattananukul, Y. Hachiya, S. Nobukawa, M. Yamaguchi, Selective localization of
12 carbon nanotubes in PC/PET blends, *Polym. Comp.* 2015; in press.
13 DOI: 10.1002/pc.23672