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Effects of super-order molecular structure of polypropylene component on mechanical properties for polypropylene/polyethylene blends and polypropylene/polystyrene blends

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Introduction

Polypropylene is most-used commodity resin has inherent characteristics, such as low specific and good moldability. However, homo polypropylene is difficult to satisfy the various demands. Therefore, polypropylene needs to be compounded with other materials e.g. polymer and inorganic filler. These multicomponent polymer blends show complicated behavior. Hence, it is important to obtain knowledge for binary blends with the purpose of guidelines for materials design of PP based blends.

Blends samples

Raw Materials

PP used in this study is the product of Japan Polypropylene Co. Ltd, with molecular weight of $M_w = 26 \times 10^4$. Two grade of HDPE with molecular weight of $M_w = 10 \times 10^4$, 5.2×10^4 , was from Tosoh Co. Ltd. sPS resin manufactured by Idemitsu Kosan Co. Ltd, Japan was used in this study. The weight average molecular weight, determined by GPC, is 19.7×10^4 .

PP/HDPE Blends preparation

Blends of different compositions of PP and HDPE were prepared by melt mixing in a 60ml batch Plastomill at 473 K and 30 rpm for 3 min. After mixing, the samples were compression molded into sheets with a hydraulic press at 503 K and 10 MPa pressure for 5min and cooled to 373 K or 195 K. Various measurements specimens were cut from the molded sheets.

PP/sPS Blends preparation

Blends of different compositions of PP and sPS were prepared by melt mixing in a 60ml batch Plastomill at 553 K and 60 rpm for 5 min. The blends were stabilized with

0.5 wt% antioxidants 1010 and 168. After mixing, the samples were compression molded into sheets with a hydraulic press at 563 K and 10 MPa pressure for 5min and cooled to 373 K or 273 K. Various measurements specimens were punched out from the molded sheets.

Polymer characterization

Density measurement

Density of samples was measured by balance. In Fig.1(a) and (b), the density of various blends is plotted against the weight percentage of HDPE and sPS. The solid line in this figure represents values obtained with the assumption that the volume of a blend is the sum of the volumes of the two components. The experimental values coincide fairly well with the estimates ones. However, only Pure PP's density with quench temperature at 195 K didn't coincide the calculated value. Pure PP having Low-density denotes to form the mesomorphic phase.

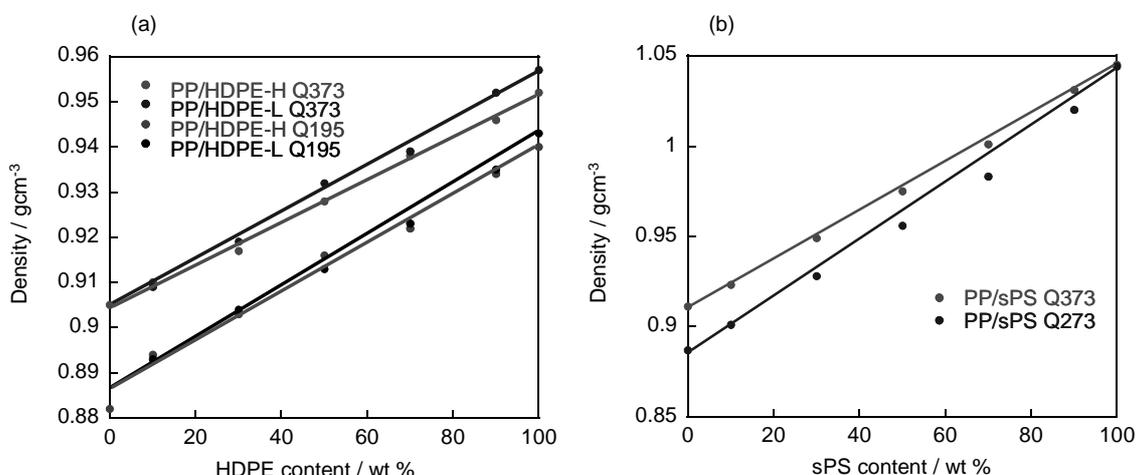


Fig. 1 The dependence of density at 293K for (a)PP/HDPE blends on HDPE content and (b) PP/sPS blends on sPS content

Result of density measurement coincide with both crystal condition was examined by wide angle X-ray diffraction and degrees of crystallinity calculated by differential scanning calorimetry.

Measurements

Mechanical properties of PP/HDPE blends

Stress-strain behavior in uniaxial tension was measured using tabletop type universal testing machine. The sample specimens were cut with a dumb-bell shape in which the gauge length is 9.53 mm. The tensile strain was calculated from the ratio of the increment of the length between clamps to the to the initial gauge length. The tensile stress was determined from dividing the tensile load by the initial cross section. Crosshead speed was 20 mm/min at 296 K.

Mechanical properties of PP/sPS blends

Tensile test performed at 293 K using dumb-bell specimens on tabletop type universal testing machine at a crosshead speed of 10 mm/min.

Rheo-optical techniques

Rheo-optical techniques afford information helpful for interpreting the stress-strain curves for crystalline polymers such as polypropylene and high density of polyethylene. When this technique is applied to blends composed of different component, it is possible to determine the molecular orientation of each component from the diachronic ratios of absorption bands characteristic to each component. In this study, infrared diachronic ratios were measured simultaneously with stress and strain during stretching the film specimen at a constant rate of elongate. Tensile test was measured at 293 K and at a crosshead speed of 1 mm/min. To determine the orientation function of molecular chains and crystal axis, we used the diachronic ratios D which can be determine by $A_{//}/A_{\perp}$ where $A_{//}$ and A_{\perp} denote the area absorbencies measured for the radiation whole electric vectors are parallel and perpendicular to the stretching direction, respectively.

The Hermans orientation function of molecular chain (or crystal axis) F is related to the diachronic ratio by equation

$$F = \frac{2}{3\cos^2 \theta} \cdot \frac{D-1}{D+2} = c \cdot \frac{D-1}{D+2} \quad (1)$$

where θ represents the angle between the stretching direction and helix chain axis for polypropylene. In this work, the crystalline orientation functions have been determined for 841 and 998 cm^{-1} absorption bands.

Result and Discussion

Fig. 2 (a) and (b) show that dependence of Young's modulus (a) on HDPE content for PP/HDPE blends and (b) on sPS content for PP/sPS blends. HDPE-H and HDPE-L indicate high molecular weight of HDPE and low molecular weight of HDPE.

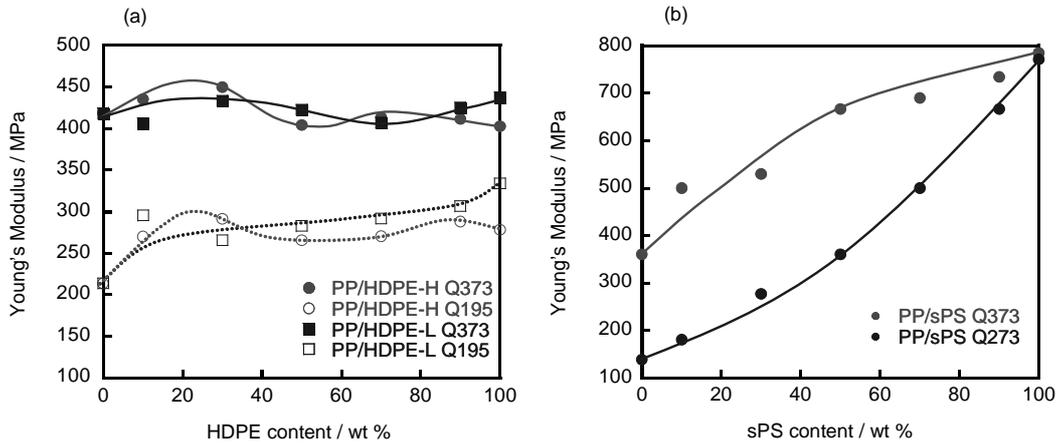


Fig.2 Dependent of Young's modulus (a) on HDPE content for PP/HDPE blends and (b) on sPS content for PP/sPS blends

As shown in Fig. 2 (a) and (b), Young's modulus of PP/HDPE blends and PP/sPS blends depends on quench temperature. This is thought to be aftereffects of super-order structure and degrees of crystallinity. Additionally, phase separation structure for HDPE and sPS content effect on Young's modulus.

Fig. 3 (a) and (b) show that dependence of Toughness (a) on HDPE content for PP/HDPE blends and (b) on sPS content for PP/sPS blends.

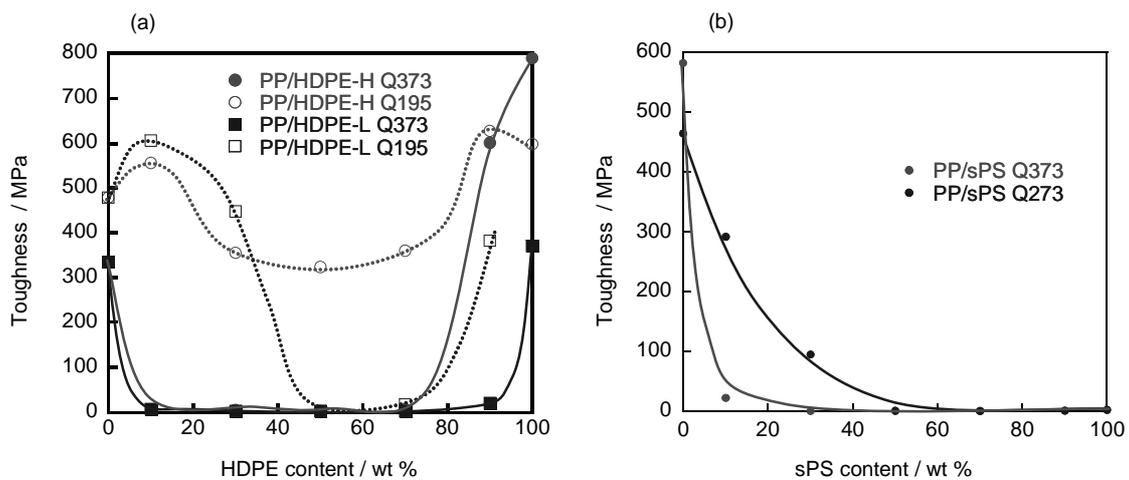


Fig.3 Dependent of Toughness (a) on HDPE content for PP/HDPE blends and (b) on sPS content for PP/sPS blends

As shown in Fig. 3 (a) and (b), Toughness for blends based PP matrix of quench

temperature at 195 K or 273 K was found higher than blends sample of quench temperature at 373 K. Then regardless of HDPE molecular weight, Toughness depends solely on quench temperature. But Toughness for blends based HDPE matrix was influenced by molecular weight of HDPE. This is thought to be aftereffects of dispersion state depending on the polymer matrix.

Fig. 4 (a) and (b) shows that orientation function for c crystallographic axis of polypropylene components for (a) PP/HDPE-H blends and (b) PP/sPS blends plotted against strain.

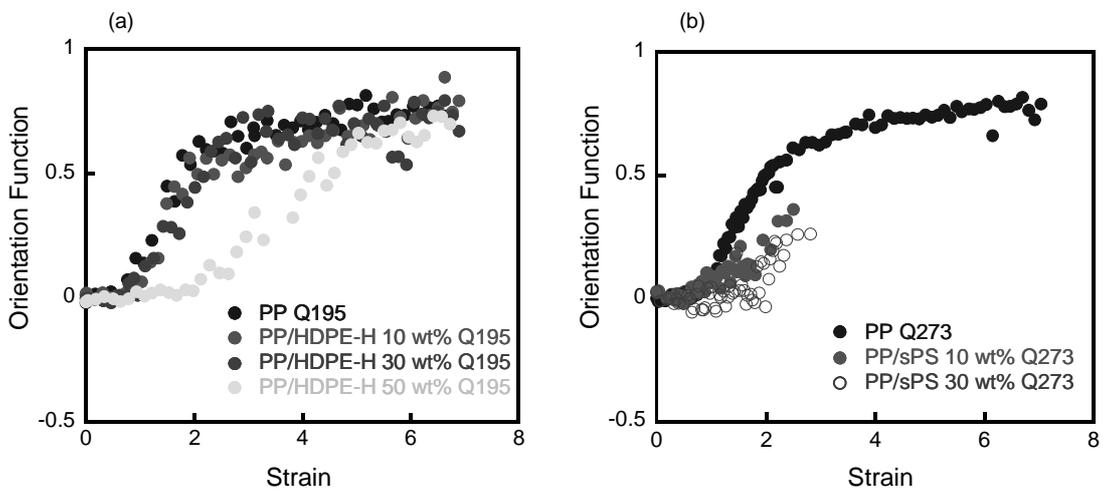


Fig.4 Orientation function for c crystallographic axis of polypropylene components for (a) PP/HDPE-H blends and (b) PP/sPS blends plotted against strain

As shown in Fig. 4 (a), only PP/HDPE-H blend 50 wt% sample has different behavior of orientation function. This is thought to be aftereffects of inhibition by phase separation structure. In the case of Fig.4 (b), orientation function for c crystallographic axis of polypropylene components inhibition by addition of sPS. For this reason, PP matrix needs greater stress in order that stress concentration as comparing with homo PP.

Conclusion

Young's modulus depends on degrees of crystallinity. Toughness of PP based blends change with higher-order structure and amount of blending polymer with polypropylene. Orientation function for c crystallographic axis of polypropylene components is influenced by phase separation structure, dispersed domain size and

modulus of elasticity for additive polymer. In this study, guidelines for materials design of PP based blends were obtained by these knowledge to effect of super-order molecular structure of polypropylene component on mechanical properties for PP based blends.

References

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