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Description	



Growth of single-crystalline zirconium diboride thin film on sapphire

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Conducting and reflecting thin film of ZrB_2 , which has lattice mismatch of only 0.6% to GaN, was grown epitaxially on sapphire substrate [α -Al₂O₃(0001)] via thermal decomposition of $Zr(BH_4)_4$. In situ reflection high energy electron diffraction and ex situ x-ray diffraction analyses indicate that the epitaxial relationship is singular, i.e., $ZrB_2[0001] \| Al_2O_3[0001]$ and $ZrB_2[11\overline{2}0] \| Al_2O_3[10\overline{1}0]$. X-ray photoelectron spectroscopy and scanning tunneling microscopy revealed that the oxide-free surface could be recovered by heating the film at approximately 750 °C under ultrahigh vacuum, which demonstrates its suitability as a template for the growth of nitride semiconductors. © 2009 American Institute of Physics. [doi:10.1063/1.3226881]

I. INTRODUCTION

Group-III nitride (III-N) semiconductors with direct band gap are now the dominating materials for shortwavelength optoelectronic devices, such as light-emitting diodes, laser diodes, and photodetectors. Sapphire has been the most commonly used substrate for the epitaxial growth of wurtzitic III-N, but since sapphire has large lattice mismatch (16%) and thermal expansion coefficient mismatch (27%) to GaN, buffer layers such as AlN (Ref. 2) or GaN (Ref. 3) grown at low temperature were crucial to achieve high quality III-N film. Recently, zirconium diboride (ZrB₂), which has a hexagonal AlB₂ structure, has been proposed as a lattice- (ZrB₂:a=3.169 Å, GaN:a=3.189 Å, 0.6% mismatch) and thermal expansion coefficient-matched $(ZrB_2:5.9\times10^{-6} \text{ K}^{-1}, GaN:5.6\times10^{-6} \text{ K}^{-1} \text{ along } [10\overline{1}0]$ direction, 5.3% mismatch) and conductive (4.6 $\mu\Omega$ cm) substrate for III-N growth. ZrB₂ is a very stable compound with high hardness and high melting temperature (3220 °C), which are desirable for III-N growth carried out under high growth temperature. If single-crystalline ZrB2 layer could be grown on sapphire, the layer can serve as a lattice-matched buffer layer as well as thermally stable electrode. ZrB2 thin film has been studied using various vapor deposition techniques, 5-7 because of its high expectation not only as wear-, heat-, and corrosion-resistant coatings but also as diffusion barrier⁵ and thermally stable electrode in electronic devices, ⁶ especially, in high power devices. ⁷ These films were amorphous or polycrystalline, but recently, ZrB2 has been epitaxially grown on Si(111) through 6:5 magic mismatch using thermal decomposition of Zr(BH₄)₄. GaN was grown on this single-crystalline ZrB2 film, and it was demonstrated that the lattice-matched growth template also acts as a metallic reflective layer, which prevents the loss of emitted light to silicon substrate. In this letter, we demonstrate that epitaxial ZrB2 layer with single crystal orientation, high

II. GROWTH PROCESS

 ${\rm ZrB_2}$ growth was carried out in a UHV system with base pressure of $1.0\times10^{-8}\,$ Pa. The flow of ${\rm Zr(BH_4)_4}$ vapor was adjusted by a leak valve, and delivered through a gas inlet capillary tube with inner diameter of 2.18 mm. The end of capillary was positioned 40 mm from the substrate. In order to grow epitaxial ${\rm ZrB_2}$ film, the pressure of ${\rm Zr(BH_4)_4}$ in the growth chamber was maintained as low as $7.0\times10^{-5}\,$ Pa to sustain a slow growth rate of approximately 0.3 nm/min. 10-mm-square and 0.5-mm-thick c-plane sapphire single crystals, which were pretreated at high temperature for atomically flat terraces, were used as substrates.

We first describe the detail growth process of ZrB₂ epilayer based on in situ reflection high-energy electron diffraction (RHEED) observation. The evolution of RHEED pattern during the growth, shown in Fig. 1, was acquired for a sample grown at 1000 °C. The evolution was common among films grown with substrate temperature between 935 and 1040 °C. Figure 1(a) shows a set of RHEED patterns from sapphire substrate with electron beam parallel to $[10\overline{10}]$ (left image) and $[11\overline{2}0]$ (right image) directions. The streaks indicate that the substrate is atomically smooth. After a few minutes after introducing Zr(BH₄)₄, spots corresponding to hexagonal ZrB₂(0001) appeared with an in-plane epitaxial relationship of $ZrB_2[11\overline{2}0]\|Al_2O_3[10\overline{1}0]$, which was also verified by ex situ x-ray diffraction (XRD) and will be discussed later. After the nucleation and growth in threedimensional mode, the growth mode changes into twodimensional, i.e., spotty pattern converts into streaky [Figs. 1(b) and 1(c)] pattern. As a result, we were able to observe a well-defined $(\sqrt{3} \times \sqrt{3})$ reconstruction [Fig. 1(d)], which maintains throughout the growth process, and also after terminating the growth at 60 min and cooling down the sample to room temperature. This $(\sqrt{3} \times \sqrt{3})$ surface reconstruction is different from the ones observed during the growth of ZrB_2 film on Si(111), $ZrB_2(0001)$ -(1×1) and (2×2), sug-

crystallinity, and high conductivity could be grown on sapphire substrate using the thermal decomposition of Zr(BH₄)₄, and that the oxide layer formed by exposing the film to air can be removed by annealing in ultrahigh vacuum (UHV).

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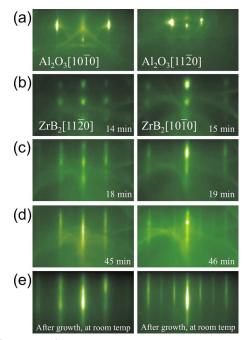


FIG. 1. (Color online) RHEED pattern evolution during the growth of ZrB2 thin film on sapphire c-plane. Patterns in the same column were acquired at the same sample rotational angle. (a) Patterns along [10 $\overline{10}$] (the left panel) and [11 $\overline{20}$] (the right panel) electron-beam azimuths of the clean Al₂O₃(0001) substrate prior to ZrB₂ growth. Patterns observed (b)–(d) during growth, and (e) after terminating the growth and cooling down to room temperature. The patterns in the left and the right columns of (b)–(e) corresponds to ZrB₂(0001) pattern with electron-beam azimuth along [11 $\overline{20}$] and [10 $\overline{10}$], respectively. A well-defined ($\sqrt{3} \times \sqrt{3}$) reconstruction is observed during and after growth.

gesting that these reconstructions are caused by different surface-segregated substrate-related elements. The observed growth rate of ZrB2 on sapphire was an order of magnitude higher than that on Si(111) under the same $Zr(BH_4)_4$ flux and substrate temperature. Similar increase in growth rate was observed for HfB2 growth from the thermal decomposition of Hf(BH₄)₄ on molybdenum and sapphire compared to that on silicon substrate. 10 We consider that the difference in growth rates of ZrB2 on sapphire and on Si(111) is due to the existence or lack of substrate-related elements, such as oxygen or silicon, which seems to affect the chemical reaction greatly. Under high growth rate, ZrB2 grew on sapphire with random crystal orientation, so in order to grow ZrB₂ on sapphire with single crystal orientation, it is necessary to reduce Zr(BH₄)₄ flux by increasing capillary distance and/or reducing the pressure in comparison to the optimum ZrB2 growth condition on Si(111).

III. STRUCTURAL AND ELECTRICAL CHARACTERIZATIONS

For the film grown for 60 min at 950 °C, thickness of 20 nm and B to Zr ratio of 2.0 with an accuracy of 5%, was measured using Rutherford backscattering spectroscopy. The crystalline quality and crystallographic orientation of the film was characterized by high-resolution XRD system (X'Pert PRO MRD, PANalytical) using Cu $K\alpha_1$ radiation. Figure 2(a) shows θ -2 θ scan of the film. No other peak was observed besides (000l) peaks of sapphire substrate and ZrB₂

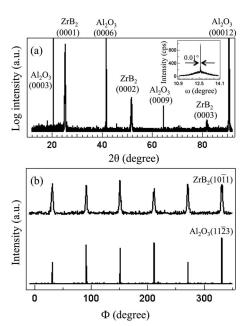


FIG. 2. (a) XRD $(\theta$ - $2\theta)$ scan of the ZrB₂/Al₂O₃(0001) using triple axis optics. Inset figure: The rocking curve of the (0001) reflection of ZrB₂. (b) Φ scan of (10 $\overline{1}$ 1) reflection for ZrB₂ and (11 $\overline{2}$ 3) reflection for Al₂O₃ substrate. Peak spacing of 60° are visible for both film and substrate according to the sixfold symmetry of ZrB₂ and sapphire; the two crystals are rotated by 30° around c-axis with respect to each other.

film. This confirms the result of RHEED, that the grown ZrB₂ film is c-axis oriented, i.e., the epitaxial relationship of ZrB₂[0001]||Al₂O₃[0001]. High-resolution rocking curve scan for ZrB₂(0001) using the triple-axis optics is shown in the inset of Fig. 2(a). Full width at half-maximum (FWHM) of the peak, 0.01°, indicates good alignment of the crystallographic planes along the c-axis, which is comparable to the result of ZrB₂ film on Si(111)⁸ and polished ZrB₂ single crystal.⁴ The c-axis lattice constant of ZrB₂ was estimated to be 3.526 ± 0.001 Å. Since the c-axis lattice constant of bulk ZrB₂ is reported to be 3.530 Å,⁴ the ZrB₂ layer grown on Al₂O₃ is nearly strain-free. The in-plane orientation of ZrB₂ film respect to sapphire substrate was determined by the Φ scan. Figure 2(b) shows the Φ scan of $ZrB_2(10\overline{11})$ and $Al_2O_3(11\overline{23})$. The $Al_2O_3(11\overline{23})$ diffraction reveals a sixfold symmetry as expected due to its hexagonal structure. The ZrB₂(1011) diffraction also reveals a sixfold symmetry which demonstrates that the grown ZrB2 film has singular in-plane crystal orientation. The peak positions of the $ZrB_2(1011)$ planes are at the same azimuthal positions as the $Al_2O_3(11\overline{23})$ planes, that results in the in-plane orientation relationship of ZrB₂[1120]||Al₂O₃[1010], which is consistent with the RHEED results. The FWHM of the $ZrB_2(10\overline{1}1)$ peaks are 2.6°, which indicates good crystalline quality. Alignment of ZrB₂ and sapphire crystals provides 33% lattice mismatch, which is favorable if the growth proceeds through 3:2 magic mismatch of 0.1% between Zr-lattice and Al-lattice. But instead, we observed 30° rotation with lattice mismatch of 15%, which corresponds to 7:8 magic mismatch of 1% between Zr-lattice and O-lattice. This mismatch is comparable to that of 6:5 magic mismatch of 1% in the case of ZrB₂(0001) on Si(111).⁸ Aligned nucleation was only ob-

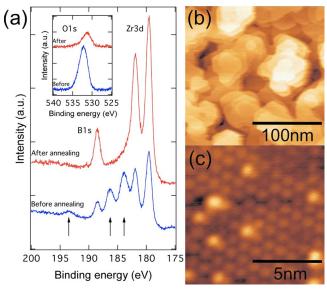


FIG. 3. (Color online) (a) XPS results for ZrB₂ film on sapphire before and after annealing at 730 °C. Oxide-related peaks of Zr 3d and B 1s, which are indicated by arrows, disappeared after the annealing. (Inset) O 1s peak intensity was reduced after annealing, but did not disappear, suggesting that the surface reconstruction is induced by oxygen. (b) $200 \times 200 \text{ nm}^2$ scanning tunneling microscopy image of oxide removed surface, showing atomically flat terraces, observed with sample bias voltage (V_s) of 1.18 V and constant tunneling current (I) of 53.4 pA. (c) $10 \times 10 \text{ nm}^2$ image of atomically resolved ZrB₂(0001)-(3×3), observed with V_s =-1.57 V and I=135.3 pA.

served at low growth temperature, below 900 $^{\circ}$ C, and together with 30 $^{\circ}$ rotation, so we consider that the hexagonal interface matching between Zr-lattice and O-lattice, and also between Al-lattice and B-lattice, results in more stable interface allowing the observed epitaxial relationship.

The electrical resistivity was measured by standard van der Pauw method at room temperature. The lowest resistivity, 20 $\mu\Omega$ cm, was measured for a stoichiometric film with thickness of 36 nm. This value is approximately four times higher than that of ZrB₂ single crystal, presumably due to the crystal size and the thickness of the film. Nevertheless, it is comparable or even superior to the reported values of 46 $\mu\Omega$ cm (Ref. 5) and 66 $\mu\Omega$ cm, ⁶ reflecting high purity and high crystallinity of the obtained film.

IV. EX SITU SURFACE CLEANING

In order to be a good template for epitaxial growth, surface oxide should be removed easily. Similar to our earlier cleaning procedure of ZrB_2 films on Si(111), we successfully cleaned ZrB_2 films on sapphire by simply heating under UHV. The samples were cut after various measurements, cleaned ultrasonically with acetone, ethanol and deionized water, and installed in our UHV scanning probe microscopy system, which is combined with x-ray photoelectron spectroscopy (XPS) and RHEED. The RHEED pattern of asloaded samples showed spots corresponding to ZrB_2 (not shown here). As-loaded sample and sample degassed at 530 °C showed Zr–O and B–O related peaks in addition to ZrB_2 -related peaks in XPS spectra measured with Mg $K\alpha$ radiation as shown in Fig. 3(a). We found that B–O peak and Zr–O peaks, which are indicated by arrows in Fig. 3(a), dis-

appeared completely after heating the sample at 730 °C for 3 h and for overnight, respectively. XPS spectra showed no further change by heating at 780 °C for more than 10 h. After oxide-related XPS peaks at Zr 3d and B 1s region disappeared, RHEED pattern changed into streaky pattern, and ZrB₂(0001)-(3×3) surface reconstruction, which is different from the pattern observed during and after growth, was observed (not shown here). Oxygen 1s peak, which is shown in the inset of Fig. 3(a), was greatly reduced after the heat treatment, but small component at the lower binding energy side remained. This result is very different from the case of ZrB₂(0001)-(2×2) observed for ZrB₂ film on Si(111), which showed no O 1s peak after the similar heat treatment. Thus, we consider that the (3×3) reconstruction in this study is induced by oxygen. Scanning tunneling microscopy was carried out on this surface, and atomically flat terraces with heights of single or multiple of c-lattice constant of ZrB_2 with $ZrB_2(0001)$ - (3×3) structure was observed, as shown in Figs. 3(b) and 3(c). The similarity in temperature and difference in surface reconstruction suggest that the oxide reduction may be intrinsic to ZrB₂ surface, but the mechanism is still under investigation.

V. CONCLUSIONS

In summary, we demonstrated that conductive and single-crystalline ZrB_2 thin film can be grown on insulating sapphire using thermal decomposition of $Zr(BH_4)_4$ under UHV. Annealing at approximately 750 °C in UHV removes oxide from the air-exposed film, and well-defined $ZrB_2(0001)$ -(3 \times 3) surface is obtained. This metallic epitaxial ZrB_2 layer on sapphire can be used not only as a growth template and/or electrode for III-N semiconductors, but also for other hexagonal wide-gap semiconductor, such as ZnO, or diborides, such as superconducting MgB_2 .

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